

Chemistry – Unit 7 Chemical Reactions

Lesson 1: Atoms

Background

Chemistry before 1900 generally focused on the study of elements. The scientific revolution in the early to mid 1800s changed that.

Historical: John Dalton's Atomic Model of Matter

The atomic theory of matter (particle theory) says that all matter consists of tiny particles which are constantly moving in a continuous motion in matter. The particles attract the neighbouring particles depending on the electrical energy they have and their distribution in other matter.

Discovery of Matter

A principle stating that matter has discrete particles is called an atomicity.

It is the total mass.

Chemical Formulae

In an atom, chemical systems undergo the various processes present in a compound and

then reaction products, which are the atoms that produce the compound.

Chemical Formulae

Chemical formulae are used to represent the chemical composition of a compound.

Chemical formulae of compounds are given by the number of atoms of each element.

It is

Chemical formulae are used to represent the chemical composition of a compound.

A chemical formula is a symbol that represents a compound.

It is

Procedure

1. The process starts with the reaction of the reactants. The reactants are placed in a closed system. They are allowed to react in the presence of the product. The reactants are placed in a closed system. They are allowed to react in the presence of the product. The reactants are placed in a closed system. They are allowed to react in the presence of the product.
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Modeling Chemistry Unit 7 Rearranging Atoms Key

Christian G. Meyer



Modeling Chemistry Unit 7 Rearranging Atoms Key:

Metals Abstracts ,1992

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