

## Chemistry Unit 8 Worksheet #1

### Examples of Binary Chemical Problems

This is a separate sheet of paper, with a complete solution to each of the problems below. Write the problems yourself on class. We can do other good class activities.

1. Calculate the amount of water of hydration attached to  $\text{MgSO}_4$  and determine how many grams of anhydrous  $\text{MgSO}_4$  is in a 1.000 g sample of hydrate given:

Masses:	$\text{MgSO}_4$	$\text{H}_2\text{O}$	$\text{MgSO}_4 \cdot \text{H}_2\text{O}$
Mass:	0.48	0.52	1.00
Volume:	0.48	0.52	1.00
Mass:	0.48	0.52	1.00

**1.000 g of  $\text{MgSO}_4$  needed**

2. In a single displacement reaction, magnesium metal reacts with hydrochloric acid to produce magnesium chloride and hydrogen gas. How many moles of hydrochloric acid were needed to completely react with a 1.0 g of magnesium?

Equation:	$\text{Mg} + 2\text{HCl} \rightarrow \text{MgCl}_2 + \text{H}_2$
Mass:	0.243 g Mg, 73 g HCl, 95 g $\text{MgCl}_2$ , 2 g $\text{H}_2$
Volume:	0.243 L $\text{MgCl}_2$ , 0.011 L $\text{H}_2$ , 0.243 L $\text{MgCl}_2$ , 0.011 L $\text{H}_2$
Mass:	0.243 g, 73 g, 95 g, 2 g

**0.041666667 moles**

3. Ethane,  $\text{C}_2\text{H}_6$ , reacts with oxygen gas to produce carbon dioxide gas and water vapor. What amount of oxygen gas is required to react with 1.00 mole of ethane?

Equation:	$\text{C}_2\text{H}_6 + 7/2 \text{O}_2 \rightarrow 2\text{CO}_2 + 3\text{H}_2\text{O}$
Mass:	30 g $\text{C}_2\text{H}_6$ , 70 g $\text{O}_2$ , 56 g $\text{CO}_2$ , 54 g $\text{H}_2\text{O}$
Volume:	0.0001 L $\text{C}_2\text{H}_6$ , 0.0001 L $\text{O}_2$ , 0.0001 L $\text{CO}_2$ , 0.0001 L $\text{H}_2\text{O}$
Mass:	0.0001 g, 0.0001 g, 0.0001 g, 0.0001 g

**3.500 moles  $\text{O}_2$  needed to react with 1.00 mole  $\text{C}_2\text{H}_6$  needed**

4. Determine the amount of water attached to a 1.00 g of a 1.00 g of hydrate. Assume that the anhydrous substance reacts with the following equation:  
 $\text{MgSO}_4 \cdot x\text{H}_2\text{O} + \text{Na}_2\text{CO}_3 \rightarrow \text{MgCO}_3 + x\text{Na}_2\text{SO}_4$

Equation:	$\text{MgSO}_4 \cdot x\text{H}_2\text{O} + \text{Na}_2\text{CO}_3 \rightarrow \text{MgCO}_3 + x\text{Na}_2\text{SO}_4$
Mass:	0.0001 g $\text{MgSO}_4$ , 0.0001 g $\text{Na}_2\text{CO}_3$ , 0.0001 g $\text{MgCO}_3$ , 0.0001 g $\text{Na}_2\text{SO}_4$
Volume:	0.0001 L $\text{MgSO}_4$ , 0.0001 L $\text{Na}_2\text{CO}_3$ , 0.0001 L $\text{MgCO}_3$ , 0.0001 L $\text{Na}_2\text{SO}_4$
Mass:	0.0001 g, 0.0001 g, 0.0001 g, 0.0001 g

**0.0001 g of water needed to react with 1.00 g of  $\text{MgSO}_4$**

# Modeling Chemistry Unit 8 3

**Neil I. Burk, Charles Edwin Bishop, Mike  
Lowe**

### **Modeling Chemistry Unit 8 3:**

**Motor Vehicle-related Air Toxics Study**, 1993      **Comprehensive Inorganic Chemistry II**, 2013-07-23

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Murray R. Gray, 2015-04-17. The emphasis throughout is to link the fundamentals of the molecules through to the economic drivers for the industry because this combination determines the technology used for processing. From the Introduction: The high demand for quality petroleum products necessitates ongoing innovation in the science and engineering underlying oilsands extraction and upgrading. Beginning with a thorough grounding in the composition, fluid properties, reaction behaviour, and economics of bitumen and heavy oil, Murray Gray then delves into current processing technologies, particularly those used at full commercial scale. The tables of data on composition, yield, and behaviour of oilsands, bitumen, and heavy oil fractions are extensive. Though the focus is on bitumen from Alberta's oilsands, the largest resource in the world, the science applies to upgrading of heavy oil and petroleum residue feeds worldwide. Upgrading Oilsands Bitumen and Heavy Oil lays out the current best practice for engineers and scientists in the oilsands and refining industries, government personnel, academics, and students.      *Inventory of Federal Energy-related Environment and Safety Research for FY 1979*, 1980      **Wetlands in**

**Tooele Valley, Utah** Neil I. Burk, Charles Edwin Bishop, Mike Lowe, 2005 This study indicates that wetlands in Tooele Valley are endangered The threats posed are from drought and increased development due to population growth which could dramatically affect the amount of water the wetlands receive sticker on back of disc cover *Inventory of Federal Energy-related Environment and Safety Research for ...* ,1980 *College of Fine Arts Catalogue* Carnegie Institute of Technology. College of Fine Arts,1922 **Growing Up with Science** Cavendish Square Publishing LLC,2006 Volume fifteen of a seventeen volume alphabetically arranged encyclopedia contains approximately five hundred articles introducing key aspects of science and technology **Timetable** University of Illinois at Urbana-Champaign,1997 **General Catalogue** Carnegie Institute of Technology,1915 Cumulated Index Medicus ,1991 **The Summer Session ...** Washington University (Saint Louis, Mo.),1927 **Report on the Education of the Architect in the United States of America** Robert Atkinson,1922 *Publications* Washington University (Saint Louis, Mo.),1927 *British Books in Print* ,1985 General Catalog San Jose State University,1925 **Review of the Student Work** Carnegie Institute of Technology. Department of Architecture,1923 Annual Report of the President of the University on Behalf of the Regents to His Excellency the Governor of the State of California University of California, Berkeley,1902 **Biennial Report of the President of the University on Behalf of the Board of Regents** University of California (System),1904 Biennial Report of the President of the University on Behalf of the Regents .. University of California (System). Regents,1908

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## **Table of Contents Modeling Chemistry Unit 8 3**

1. Understanding the eBook Modeling Chemistry Unit 8 3
  - The Rise of Digital Reading Modeling Chemistry Unit 8 3
  - Advantages of eBooks Over Traditional Books
2. Identifying Modeling Chemistry Unit 8 3
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Modeling Chemistry Unit 8 3
  - User-Friendly Interface
4. Exploring eBook Recommendations from Modeling Chemistry Unit 8 3
  - Personalized Recommendations
  - Modeling Chemistry Unit 8 3 User Reviews and Ratings
  - Modeling Chemistry Unit 8 3 and Bestseller Lists

5. Accessing Modeling Chemistry Unit 8 3 Free and Paid eBooks
  - Modeling Chemistry Unit 8 3 Public Domain eBooks
  - Modeling Chemistry Unit 8 3 eBook Subscription Services
  - Modeling Chemistry Unit 8 3 Budget-Friendly Options
6. Navigating Modeling Chemistry Unit 8 3 eBook Formats
  - ePub, PDF, MOBI, and More
  - Modeling Chemistry Unit 8 3 Compatibility with Devices
  - Modeling Chemistry Unit 8 3 Enhanced eBook Features
7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Modeling Chemistry Unit 8 3
  - Highlighting and Note-Taking Modeling Chemistry Unit 8 3
  - Interactive Elements Modeling Chemistry Unit 8 3
8. Staying Engaged with Modeling Chemistry Unit 8 3
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Modeling Chemistry Unit 8 3
9. Balancing eBooks and Physical Books Modeling Chemistry Unit 8 3
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Modeling Chemistry Unit 8 3
10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
11. Cultivating a Reading Routine Modeling Chemistry Unit 8 3
  - Setting Reading Goals Modeling Chemistry Unit 8 3
  - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Modeling Chemistry Unit 8 3
  - Fact-Checking eBook Content of Modeling Chemistry Unit 8 3
  - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
14. Embracing eBook Trends
- Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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