

## General Chemistry – Unit 3 Review

To prepare to do well on the Unit 3 test, you should assemble your notes, the 4 worksheets, labs and activity notes, and the quizzes and review them, preferably in a small group where you can draw from each other's understanding. Here are the key points you should know.

### Energy

Think of energy as a quantity that is always involved when there is a *change* in the state of matter. When a substance gets hotter or colder or changes phase, energy is either transferred into or out of the system. The two key ways energy is stored is **kinetic** (due to the motion of the particles) and **interaction** (due to attractions between the particles). Remember that attractions *lower* the energy state, so one must *add* energy to a system to pull particles apart. The three ways that energy is transferred is by heating (Q), working (W) and radiating (R); this course focuses on Q. You will be expected to be able to

1. Draw energy bar graphs to account for energy storage and transfer in all sorts of changes. Make up a sample situation and sketch the bar graph. (review ws 1 and 2, quiz)



# Modeling Chemistry Unit 3 Review

**Suresh K. Sharma**



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