

Chemistry – Unit 7 Review: Chemical Reactions

- ___ 1. Numbers placed in front of formulas in an equation are called
- ___ 2. New substances formed in a chemical reaction are called
- ___ 3. The total number of atoms represented by the formula $\text{Fe}_3(\text{PO}_4)_2$ is
- ___ 4. The symbol that indicates a substance dissolved in water is
- ___ 5. Two white powders are mixed in a flask. The bottom of the flask feels cooler. This an example of what kind of reaction?
- ___ 6. The charge on lead in PbSO_4 is



- _____ 7. How many more D atoms should be on the right side of the equation?
- _____ 8. What is the formula of the missing product molecule?
- _____ 9. What are particles on the left side of the arrow called?
10. Write the completed chemical equation for the reaction pictured.

Balance the following reactions.



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