

Chemistry – Unit 7 Review: Chemical Reactions

- ___ 1. Numbers placed in front of formulas in an equation are called
- ___ 2. New substances formed in a chemical reaction are called
- ___ 3. The total number of atoms represented by the formula $\text{Fe}_3(\text{PO}_4)_2$ is
- ___ 4. The symbol that indicates a substance dissolved in water is
- ___ 5. Two white powders are mixed in a flask. The bottom of the flask feels cooler. This an example of what kind of reaction?
- ___ 6. The charge on lead in PbSO_4 is



- _____ 7. How many more D atoms should be on the right side of the equation?
- _____ 8. What is the formula of the missing product molecule?
- _____ 9. What are particles on the left side of the arrow called?
10. Write the completed chemical equation for the reaction pictured.

Balance the following reactions.



Modeling Chemistry Unit 7 Answer

Brendan G. Carr



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Chemistry in the Community (Enhanced Core Four) American Chemical Society,2006-02-15 Energy Research Abstracts ,1982 *Study Guide and Solutions Manual for McMurry's Organic Chemistry, Fifth Edition* Susan McMurry,2000 Provides answers and explanations to all in text and end of chapter exercises Also includes summaries of name reactions functional group synthesis and reactions lists of reagents and abbreviations and articles on topics ranging from infrared absorption frequencies to the Nobel Prize winners in Chemistry This edition now includes all new artwork expanded in text problems summary quizzes approximately every three chapters more detailed explanations in solutions and chapter outlines

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