

Isotopes Practice Set

- What does "atomic number" mean? Z is the number of protons in the nucleus.
- What does the number next to isotopes mean? A is the mass number, the sum of protons and neutrons.
- How do you find isotopes? $A - Z = N$ (number of neutrons)

For each of the following isotopes, write the number of protons, neutrons, and electrons.

	Chlorine-35	Chlorine-37
# of protons	17	17
# of neutrons	18	20
# of electrons	17	17

	Magnesium-24	Magnesium-26
# of protons	12	12
# of neutrons	12	14
# of electrons	12	12

	Carbon-12	Carbon-13
# of protons	6	6
# of neutrons	6	7
# of electrons	6	6

	Sulfur-32	Sulfur-34
# of protons	16	16
# of neutrons	16	18
# of electrons	16	16

Fill in the isotopes names including atomic numbers, including atomic numbers from the chart. Use your periodic table and the information provided.

	Isotope: $^{35}_{17}\text{Cl}$	Isotope: $^{37}_{17}\text{Cl}$
# of protons	17	17
# of neutrons	18	20
# of electrons	17	17

	Isotope: $^{12}_{6}\text{C}$	Isotope: $^{13}_{6}\text{C}$
# of protons	6	6
# of neutrons	6	7
# of electrons	6	6

	Isotope: $^{24}_{12}\text{Mg}$	Isotope: $^{26}_{12}\text{Mg}$
# of protons	12	12
# of neutrons	12	14
# of electrons	12	12

	Isotope: $^{32}_{16}\text{S}$	Isotope: $^{34}_{16}\text{S}$
# of protons	16	16
# of neutrons	16	18
# of electrons	16	16

Isotope Practice Set Answers

Huangqi Zhang



Isotope Practice Set Answers:

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The Publishers' Trade List Annual ,1968

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Safe Handling of Radioactive Isotopes in Medical Practice Edith Hinkley Quimby,1960

The Michigan Technic ,1950

The Skipper ,1960

Abstracts of Papers American Chemical Society. Meeting,American Chemical Society,1959

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