

- (a) $(-2a^2b^3)^2 = 4a^4b^6$
- (b) $(-2a^2b^3)^3 = -8a^6b^9$
- (c) $(-2a^2b^3)^4 = 16a^8b^{12}$
- (d) $(-2a^2b^3)^5 = -32a^{10}b^{15}$
- (e) $(-2a^2b^3)^6 = 64a^{12}b^{18}$

EX.2 Simplify completely:

(a) $(x^2y^3)^2 = (x^2)^2 (y^3)^2 = x^4y^6$

(b)

Quantitative Aptitude Questions & Answers

(c)

(d) $(-2 a^2 b^3)^4 = (-2)^4 a^{2 \times 4} b^{3 \times 4} = 16 a^8 b^{12}$

(e) $(-2a^2 b^3)^3 = (-2)^3 a^{2 \times 3} b^{3 \times 3} = -8 a^6 b^9$

(f) $(-2x^2 y^3)^4 = (-2)^4 x^{2 \times 4} y^{3 \times 4} = 16 x^8 y^{12}$

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web october 3 2017 abstract the techniques used in this lab are quantitative transfer and vacuum filtration with the reaction of 8 001 grams of copper ii sulfate CuSO_4 and 2 0153 grams of iron powder Fe the goal of this experiment was to determine the product of copper ii sulfate with iron

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web procedure 1 find the mass of a clean dry 100 ml beaker and record 2 measure out 8 00 grams of copper ii sulfate onto a piece of filter paper taking the filter paper s mass into account add these to the beaker 3 measure 50 0 ml of water in the cylinder and add it to the beaker 4

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