

## Like Dissolves Like Practice Problems

Using your knowledge of the "like dissolves like" principle of polarity, determine which component will dissolve best in each of the following solvents. Explain your choice for each solvent and the determination.

1) Solvent: methanol (CH<sub>3</sub>OH)

Components: carbon tetrachloride and oxygen dichloride

Both oxygen dichloride and methanol are polar, while carbon tetrachloride is not.

2) Solvent: carbon dioxide

Components: methanol (CH<sub>3</sub>OH) and carbon dioxide

While both components are polar, methanol is considerably more polar than carbon dioxide (both components always have full charge separation, while carbon dioxide, as neutral does not), so not.

3) Solvent: water

Components: carbon tetrachloride and ammonia

Both water and ammonia are polar. Carbon tetrachloride is not.

4) Solvent: hexane (C<sub>6</sub>H<sub>14</sub>)

Components: water and methanol

Both water and methanol are polar. Methanol is not.

# Like Dissolves Like Practice Answer Key

**SB Merriam**



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**Chemistry** John A. Olmsted, Gregory M. Williams, Robert C. Burk, 2020-12-28 Olmsted Chemistry 4th Canadian Edition is a rigorous yet accessible general chemistry textbook crafted specifically for Canadian students and professors Featuring SI units IUPAC standards and Canadian case studies this edition aligns closely with curricula across Canadian institutions Its clear language logical structure and engaging problem solving approach help students build confidence and competence without overwhelming them with redundancy

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**Chemical Principles of Textile Conservation** Agnes Timar-Balazsy, Dinah Eastop, 2012-09-10 Chemical Principles of Textile Conservation provides must have knowledge for conservators who do not always have a scientific background This vital book brings together from many sources the material science necessary to understand the properties deterioration and investigation of textile artefacts It also aids understanding of the chemical processes during various treatments such as cleaning humidification drying disinfestation disinfection and the use of adhesives and consolidants in conservation of historical textiles Textile conservators will now have ready access to the necessary knowledge to understand the chemistry of the objects they are asked to treat and to make informed decisions about how to preserve textiles The combination of a chemist and a conservator provides the perfect authorial team It ensures a unique dual function of the text which provides textile conservators with vital chemical knowledge and gives scientists an understanding of textile conservation necessary to direct their research The many practical examples and case studies illustrate the utility of the relatively large chemical introduction and the essential chemical information which is included The case studies many illustrated in colour range from the treatment of the Ghandis clothes high altitude flying suits and a Mary Quant raincoat to the Hungarian Coronation Mantle

**Chemistry** Trace Jordan, Neville R. Kallenbach, 2017 Chemistry The Molecules of Life offers chemical insights within the context of health pharmaceuticals and the function of biological molecules The contextualized presentation of topics gives students a broad introduction to chemistry and helps them to see the relevance of chemistry to their personal lives

**Developing Solid Oral Dosage Forms** Yihong Qiu, Yisheng Chen, Geoff G.Z. Zhang, Lawrence Yu, Rao V. Mantri, 2016-11-08 Developing Solid Oral Dosage Forms Pharmaceutical Theory and Practice

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