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GRADE 9 NATURAL SCIENCES JUNE TEST

SECTION A

QUESTION 1

1.1 Various options are provided as answers to the following questions. Choose the correct answer and write **ONLY** the LETTER(A-D) next to the QUESTION NUMBER (1.1.1-1.1.10) on your answer sheet, for example 1.1.11 D.

1.1.1 Which of the following can you NOT learn from a chemical formula?

- A Elements in the compound
- B Phase of the element
- C Ratio of atoms
- D Total number of atoms

1.1.2 Which of the following is NOT a sub-atomic particle?

- A Electrons
- B Protons
- C Neutrons
- D Nucleons

1.1.3 What is the prefix for 6 atoms?

- A Hepta-
- B Hexa-
- C Penta-
- D Tri-

1.1.4 An element has an atomic number of 30 and mass number of 65. How many nucleons will it have?

- A 30
- B 35
- C 65
- D 95

1.1.5 Element at group 15, period 2 on the periodic table...

- A Carbon
- B Nitrogen
- C Silicon
- D Oxygen

1.1.6 The following is an example of a semi-metal:

- A Silicon
- B Lithium
- C Magnesium
- D Carbon

1.1.7 Substances on the left side of a chemical equation:

- A Products
- B Reactants
- C Metals
- D Non-metals

1.1.8 Which of the following is NOT a way to prevent rust?

- A Galvanisation
- B Electrolysis
- C Paint
- D Oil or grease

1.1.9 Substance with a pH lower than 7 will be classified as:

- A Lemon Juice
- B Neutral
- C Acidic
- D Alkaline

1.1.10 An neutral substance would turn _____ with universal indicator.

- A Green
- B Violet
- C Colourless
- D Red

[10x1]

1.2 Provide the correct TERM for the following descriptions. Write **ONLY** the correct TERM next to the QUESTION NUMBER (1.2.1-1.2.7) on your answer sheet.

1.2.1 Measure of how strongly acidic or alkaline a substance is.

1.2.2 Reaction of a substance with oxygen.

1.2.3 Name for group 18 on the periodic table.

1.2.4 Process whereby compounds can be broken down using electricity.

1.2.5 H_2 , O_2 , Br_2 , I_2 .

1.2.6 Name for groups 3 to 12 on the periodic table.

1.2.7 When acids and bases react with each other.

[1x7]

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E Durkheim



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Table of Contents Grade 9 Ns June Examination Paper

1. Understanding the eBook Grade 9 Ns June Examination Paper
 - The Rise of Digital Reading Grade 9 Ns June Examination Paper
 - Advantages of eBooks Over Traditional Books
2. Identifying Grade 9 Ns June Examination Paper
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Grade 9 Ns June Examination Paper
 - User-Friendly Interface
4. Exploring eBook Recommendations from Grade 9 Ns June Examination Paper
 - Personalized Recommendations
 - Grade 9 Ns June Examination Paper User Reviews and Ratings

- Grade 9 Ns June Examination Paper and Bestseller Lists
- 5. Accessing Grade 9 Ns June Examination Paper Free and Paid eBooks
 - Grade 9 Ns June Examination Paper Public Domain eBooks
 - Grade 9 Ns June Examination Paper eBook Subscription Services
 - Grade 9 Ns June Examination Paper Budget-Friendly Options
- 6. Navigating Grade 9 Ns June Examination Paper eBook Formats
 - ePub, PDF, MOBI, and More
 - Grade 9 Ns June Examination Paper Compatibility with Devices
 - Grade 9 Ns June Examination Paper Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Grade 9 Ns June Examination Paper
 - Highlighting and Note-Taking Grade 9 Ns June Examination Paper
 - Interactive Elements Grade 9 Ns June Examination Paper
- 8. Staying Engaged with Grade 9 Ns June Examination Paper
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Grade 9 Ns June Examination Paper
- 9. Balancing eBooks and Physical Books Grade 9 Ns June Examination Paper
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Grade 9 Ns June Examination Paper
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Grade 9 Ns June Examination Paper
 - Setting Reading Goals Grade 9 Ns June Examination Paper
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Grade 9 Ns June Examination Paper
 - Fact-Checking eBook Content of Grade 9 Ns June Examination Paper
 - Distinguishing Credible Sources

13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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