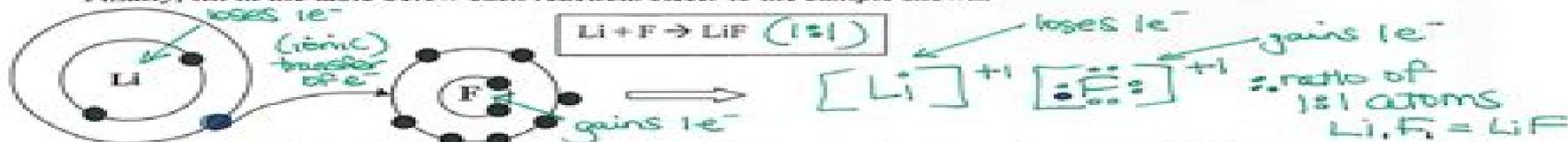


Ionic Bonding Worksheet

For each pair of elements below draw an atomic diagram showing electrons in different energy levels. Draw arrows to show where the outer electrons will go during a chemical reaction, then draw the resulting compound. Finally, fill in the table below each reaction. Refer to the sample shown.



Atoms	Valence electrons	Electron transfer from/to each atom	Ions formed in the product
Li	1ve	transfers 1ve ⁻ to F.	Li ⁺
F	7ve	accepts 1ve ⁻ from Li	F ⁻

Reactions	Atoms	Valence electrons	Electron transfer from/to each atom	Ions formed in the product
1) $\text{Li} + \text{Cl} \Rightarrow \text{LiCl}$ 	Li	1ve	donate 1e ⁻	Li ⁺
	Cl	7ve	accepts 1e ⁻	Cl ⁻
2) $\text{Ca} + \text{O} \Rightarrow \text{CaO}$ 	Ca	2ve	donates 2e ⁻	Ca ⁺²
	O	6ve	accepts 2e ⁻	O ⁻²
3) $\text{Be} + \text{F} \Rightarrow \text{BeF}_2$ 	Be	2ve	donates 2e ⁻ to each F.	Be ⁺²
	F	7ve	each F accepts 1ve each	F ⁻¹
4) $\text{Mg} + \text{S} \Rightarrow \text{MgS}$ 	Mg	2ve	donates 2e ⁻ to S	Mg ⁺²
	S	6ve	accepts 2e ⁻	S ⁻²
5) $\text{K} + \text{F} \Rightarrow \text{KF}$ 	K	1ve	donates 1ve ⁻	K ⁺¹
	F	7ve	accepts 1ve ⁻	F ⁻¹

Ionic Bonding Worksheet Answer Key

Primrose Kitten



Ionic Bonding Worksheet Answer Key:

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Drawing electron configurations

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Table of Contents Ionic Bonding Worksheet Answer Key

1. Understanding the eBook Ionic Bonding Worksheet Answer Key
 - The Rise of Digital Reading Ionic Bonding Worksheet Answer Key
 - Advantages of eBooks Over Traditional Books
2. Identifying Ionic Bonding Worksheet Answer Key
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Ionic Bonding Worksheet Answer Key
 - User-Friendly Interface
4. Exploring eBook Recommendations from Ionic Bonding Worksheet Answer Key
 - Personalized Recommendations
 - Ionic Bonding Worksheet Answer Key User Reviews and Ratings

- Ionic Bonding Worksheet Answer Key and Bestseller Lists
- 5. Accessing Ionic Bonding Worksheet Answer Key Free and Paid eBooks
 - Ionic Bonding Worksheet Answer Key Public Domain eBooks
 - Ionic Bonding Worksheet Answer Key eBook Subscription Services
 - Ionic Bonding Worksheet Answer Key Budget-Friendly Options
- 6. Navigating Ionic Bonding Worksheet Answer Key eBook Formats
 - ePub, PDF, MOBI, and More
 - Ionic Bonding Worksheet Answer Key Compatibility with Devices
 - Ionic Bonding Worksheet Answer Key Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Ionic Bonding Worksheet Answer Key
 - Highlighting and Note-Taking Ionic Bonding Worksheet Answer Key
 - Interactive Elements Ionic Bonding Worksheet Answer Key
- 8. Staying Engaged with Ionic Bonding Worksheet Answer Key
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Ionic Bonding Worksheet Answer Key
- 9. Balancing eBooks and Physical Books Ionic Bonding Worksheet Answer Key
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Ionic Bonding Worksheet Answer Key
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Ionic Bonding Worksheet Answer Key
 - Setting Reading Goals Ionic Bonding Worksheet Answer Key
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Ionic Bonding Worksheet Answer Key
 - Fact-Checking eBook Content of Ionic Bonding Worksheet Answer Key
 - Distinguishing Credible Sources

13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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