

## PART II : CHEMISTRY

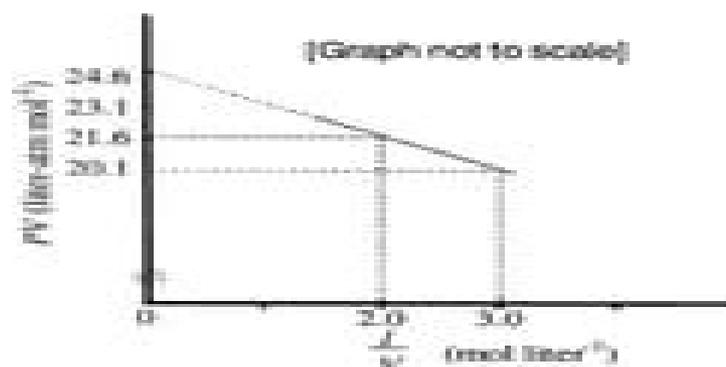
### SECTION I : Single Correct Answer Type

This section contains 10 multiple choice questions. Each question has four choices (A), (B), (C) and (D) out of which **ONLY ONE** is correct.

21. In allene ( $C_3H_2$ ), the type(s) of hybridisation of the carbon atoms is (are)  
(A)  $sp$  and  $sp^2$       (B)  $sp$  and  $sp^3$       (C) only  $sp^2$       (D)  $sp^2$  and  $sp^3$

**ANSWER : B**

22. For one mole of a van der Waals gas when  $b = 0$  and  $T = 300\text{ K}$ , the  $PV$  vs.  $V/V^2$  plot is shown below. The value of the van der Waals constant  $a$  ( $\text{atm.lit}^2\text{ mol}^{-2}$ ) is



- (A) 1.0      (B) 4.5      (C) 1.5      (D) 3.0

**ANSWER : C**

# Iit Jee 2012 Question Paper

**Terry C. Jones**



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