

# Ionic Exchange

The goal of this experiment is to understand the concept of the cation and anion exchange ion-exchange resin. It includes the following examples:  $\text{Na}^+$ ,  $\text{Ca}^{2+}$ ,  $\text{Mg}^{2+}$ . You should write down each of the changes to water.

Reverse ion-exchange resin is used in water softening. This resin has negatively charged groups (the resin) which the cations  $\text{Ca}^{2+}$  is called the "divalent ion" and  $\text{Mg}^{2+}$  is called the "monovalent ion". Ion-exchange resin is used to soften water. It is used to remove the hardness of the water. The hardness of water is caused by  $\text{Ca}^{2+}$  and  $\text{Mg}^{2+}$  ions.

## Critical Thinking Questions

1. Write the chemical formula for the cation and anion of the following salts:  $\text{NaCl}$ ,  $\text{CaCl}_2$ ,  $\text{MgCl}_2$ ,  $\text{Na}_2\text{SO}_4$ ,  $\text{CaSO}_4$ ,  $\text{MgSO}_4$ .

Caution: Do not use the following salts in your experiment:  $\text{NaCl}$ ,  $\text{CaCl}_2$ ,  $\text{MgCl}_2$ ,  $\text{Na}_2\text{SO}_4$ ,  $\text{CaSO}_4$ ,  $\text{MgSO}_4$ .

## Write equations for the following and show products.

The following are used to soften water. Write the chemical formula for each of the following. Write the chemical formula for each of the products. Write the chemical formula for each of the reactants. Write the chemical formula for each of the products. Write the chemical formula for each of the reactants. Write the chemical formula for each of the products.

- $\text{Na}^+$  ion-exchange resin and  $\text{Ca}^{2+}$  ion-exchange resin
- $\text{Mg}^{2+}$  ion-exchange resin and  $\text{Ca}^{2+}$  ion-exchange resin
- $\text{Ca}^{2+}$  ion-exchange resin and  $\text{Mg}^{2+}$  ion-exchange resin
- $\text{Mg}^{2+}$  ion-exchange resin and  $\text{Na}^+$  ion-exchange resin

The chemical formula of the following is used to soften water. Write the chemical formula for each of the following. Write the chemical formula for each of the products. Write the chemical formula for each of the reactants. Write the chemical formula for each of the products.

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2. Write the chemical formula for the cation and anion of the following salts:  $\text{NaCl}$ ,  $\text{CaCl}_2$ ,  $\text{MgCl}_2$ ,  $\text{Na}_2\text{SO}_4$ ,  $\text{CaSO}_4$ ,  $\text{MgSO}_4$ . Write the chemical formula for the cation and anion of the following salts:  $\text{NaCl}$ ,  $\text{CaCl}_2$ ,  $\text{MgCl}_2$ ,  $\text{Na}_2\text{SO}_4$ ,  $\text{CaSO}_4$ ,  $\text{MgSO}_4$ .
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# Ionic Bonding Advanced Chemquest 20 Key

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