

## Matter Test Study Guide



**matter** = anything that has mass and takes up space.

**pure substance** = a single kind of matter that is pure, meaning that it always has a specific chemical makeup. Ex.) table salt is always NaCl and sugar is always C<sub>12</sub>H<sub>22</sub>O<sub>11</sub>

**element**= a pure substance that can't be broken down into any other substance. Ex.) oxygen, copper, aluminum, zinc, tungsten, carbon, hydrogen, helium, nitrogen

**atom**= the smallest part of an element. Ex.) one atom of oxygen



**compound**= a pure substance made of two or more elements chemically combined in a set ratio. Ex.) H<sub>2</sub>O

**molecule**= the smallest part of a compound Ex.) one molecule of H<sub>2</sub>O



**mixture** = made of two or more substances that are mixed together, but not chemically combined. Ex.) trail mix

**homogeneous**= can't see the different parts. Ex.) sugar water

**heterogeneous**= can see the different parts. Ex.) salad



**solution** = an example of a homogenous mixture

**chemical formulas**= tell you which elements and how many atoms of each are in a compound ex.) C<sub>12</sub>H<sub>22</sub>O<sub>11</sub>

**set ratio** = a specific number of atoms of each element to make a compound

**weight** = a measure of the force of gravity on an object; can change- depends on location  
Ex.) earth vs. moon

**mass** = a measure of the amount of matter in an object; **does not change**

- triple beam balance
- SI unit is the kilogram
- scientists prefer to measure by mass rather than weight because mass doesn't change



My **WEIGHT** on Earth is around 560N



My **WEIGHT** on the moon is around 90N



My **MASS** is always 56kg!

# Matter Study Guide

**SA Adler**



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