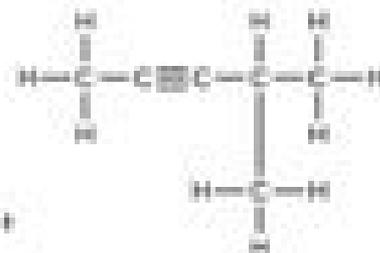
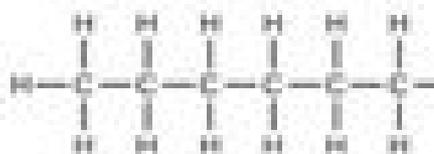


## TOPIC 1 -

# Hydrocarbons



Organic compounds containing only hydrogen and carbon atoms.

## Key Facts

- Hydrocarbons may have one or more branched chains or cyclic structures.
- The hydrogen and carbon atoms have similar electronegativities (2.20 and 2.55) making them have low solubility in polar solvents.
- Hydrocarbons have the weakest intermolecular forces (London dispersion forces).
- Weak intermolecular forces → low boiling and melting points.
- 3 main groups of hydrocarbon
  - Alkanes ⇒ single carbon-carbon bonds (saturated hydrocarbons)
  - Alkenes ⇒ contain at least one carbon-carbon double bond (unsaturated hydrocarbons)
  - Alkynes ⇒ contain at least one carbon-carbon triple bond (unsaturated)

## 3 Main Groups of Hydrocarbon

### Alkanes

- General formula =  $\text{C}_n\text{H}_{2n+2}$
- Hydrocarbons with single

### Alkenes

- General formula =  $\text{C}_n\text{H}_{2n}$
- Hydrocarbons with at least

### Alkynes

- General formula =  $\text{C}_n\text{H}_{2n-2}$
- Hydrocarbons with at least

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**Patrick Vollmar**



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