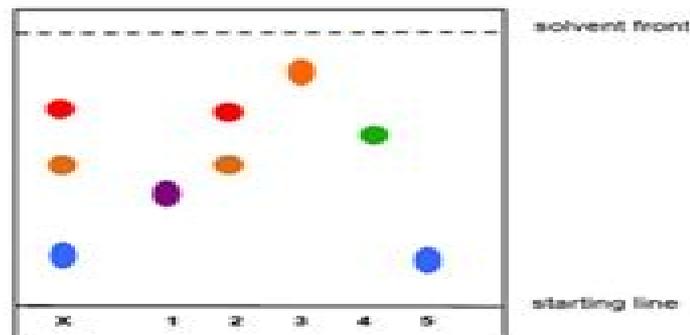




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A student analysed substance X by paper chromatography, comparing it to five other substances (1-5).

The student drew a starting line on the paper with a pencil and placed spots of the samples on the line. The paper was hung in a beaker of ethanol, with the ethanol being below the height of the samples on the line. The ethanol soaked up the paper. When it was near the top, it was removed from the beaker and the solvent front marked on the paper.



1 Why is the starting line drawn in pencil?Because ink (for example) would also dissolve in the solvent, while a line made out of pencil doesn't move.

2 Why is the level of the solvent in the beaker below the sample line?
To give an accurate measurement of the distance between the starting line and the substances tested, as well as giving the scientists more time to set everything up, without disturbing the moving and stationary phases.

3 What is the stationary phase in this experiment? ...The paper.....

4 What is the mobile phase in this experiment? The ethanol.....

5 Explain, by referring to the stationary and mobile phases, why the samples move at different speeds during the experiment.

Some samples spend more time in the mobile phase, making them move more with the solvent, while other samples spend more time in the stationary phase, making them move less and stay at the same position more, making them travel less distance overall.

6 Which of samples 1-5 are in sample X?

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