

The isotope fractionation factor

The isotope fractionation factor is defined as: $\alpha_b^a = \frac{R_a}{R_b}$

where R_A , R_B are the isotope ratios in two phases (e.g., carbonate and water, or water vapor and water)

For example, consider: $\text{H}_2\text{O}(l) \leftrightarrow \text{H}_2\text{O}(v)$ at 25°C

$$\alpha_v^l(\text{O}) = \frac{R_l}{R_v} = \frac{\left(\frac{{}^{18}\text{O}}{{}^{16}\text{O}}\right)_l}{\left(\frac{{}^{18}\text{O}}{{}^{16}\text{O}}\right)_v} = 1.0092$$

The differences in isotope ratios are relatively small and are expressed as parts per thousand (per mil) deviations from a standard.

Fractionation Formula Examples

JR Anderson



Fractionation Formula Examples:

Unveiling the Power of Verbal Beauty: An Mental Sojourn through **Fractionation Formula Examples**

In a global inundated with displays and the cacophony of instant interaction, the profound power and psychological resonance of verbal artistry usually diminish in to obscurity, eclipsed by the regular assault of noise and distractions. Yet, located within the lyrical pages of **Fractionation Formula Examples**, a fascinating function of fictional splendor that pulses with raw emotions, lies an memorable trip waiting to be embarked upon. Written by way of a virtuoso wordsmith, this mesmerizing opus books readers on a psychological odyssey, lightly exposing the latent possible and profound influence stuck within the complex web of language. Within the heart-wrenching expanse of the evocative evaluation, we can embark upon an introspective exploration of the book is main subjects, dissect its captivating publishing model, and immerse ourselves in the indelible impression it leaves upon the depths of readers souls.

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