



## DAILY LESSON PLAN IN PHYSICAL SCIENCE 12

**Grade & Section:**  
12 - ABM

**Quarter/Semester:**  
3<sup>rd</sup> Quarter, 2<sup>nd</sup> Sem

**Duration:**  
1 hour

I. OBJECTIVES	
Content Standards	The learners demonstrate an understanding of: 1. How the concept of the atom evolved from Ancient Greek to the present
Performance Standards	The learners shall be able to make a creative representation of the historical development of the atom or the chemical element in a timeline.
Learning Competency	The learners to: 1. Describe the ideas of the Ancient Greeks on the atom (S11/12PS-IIIa-b-5)
Specific Learning Outcomes	At the end of the lesson, the learners shall be able to: 1. describe the key concepts of the atomic structure; 2. identify the contributions of scientists in the development of atomic structure; and 3. appreciate the significance and application of the discovery of atomic structure in real-life.
II. CONTENT	
History of Atomic Structure	
III. LEARNING RESOURCES	
A. Materials	Projector, manila paper, felt-tip pen, laptop
B. References	Teaching Guide in SHS Physical Science pp. 34-47 <a href="https://medium.com/@intlink.edu/a-timeline-of-atomic-models-cb2607b1da85">https://medium.com/@intlink.edu/a-timeline-of-atomic-models-cb2607b1da85</a> <a href="https://www.compoundchem.com/2016/10/13/atomicmodels/">https://www.compoundchem.com/2016/10/13/atomicmodels/</a> <a href="https://www.youtube.com/watch?v=GUaAeXcbf7c">https://www.youtube.com/watch?v=GUaAeXcbf7c</a>
C. Strategies	Cooperative Learning, Differentiated Instruction
D. Approach	4 A's
IV. PROCEDURES	
a. Preliminaries	1. Prayer 2. Greetings 3. Checking of Attendance
b. Review	Have a quick review of the topic last meeting by asking questions.
Motivation	Show the illustration to the students and ask them the ff: *What is the illustration about? *Why does the atomic structure have different/changing models over time?
<h3>ATOMIC MODELS</h3>  <p>1803 JOHN DALTON      1904 J.J. THOMSON      1911 ERNEST RUTHERFORD      1913 NIELS BOHR      1926 ERWIN SCHRÖDINGER</p>	
<b>ACTIVITY</b>	Students are grouped into 7 (corresponding to the number of scientists) and are given activity sheets and materials for the activity. see ifachment)  *From your English class, what is the W and H questions? These are the questions asking what, who, when, where, why and how. Now, you will be

# Kindergarten Physical Science Lesson Plan

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