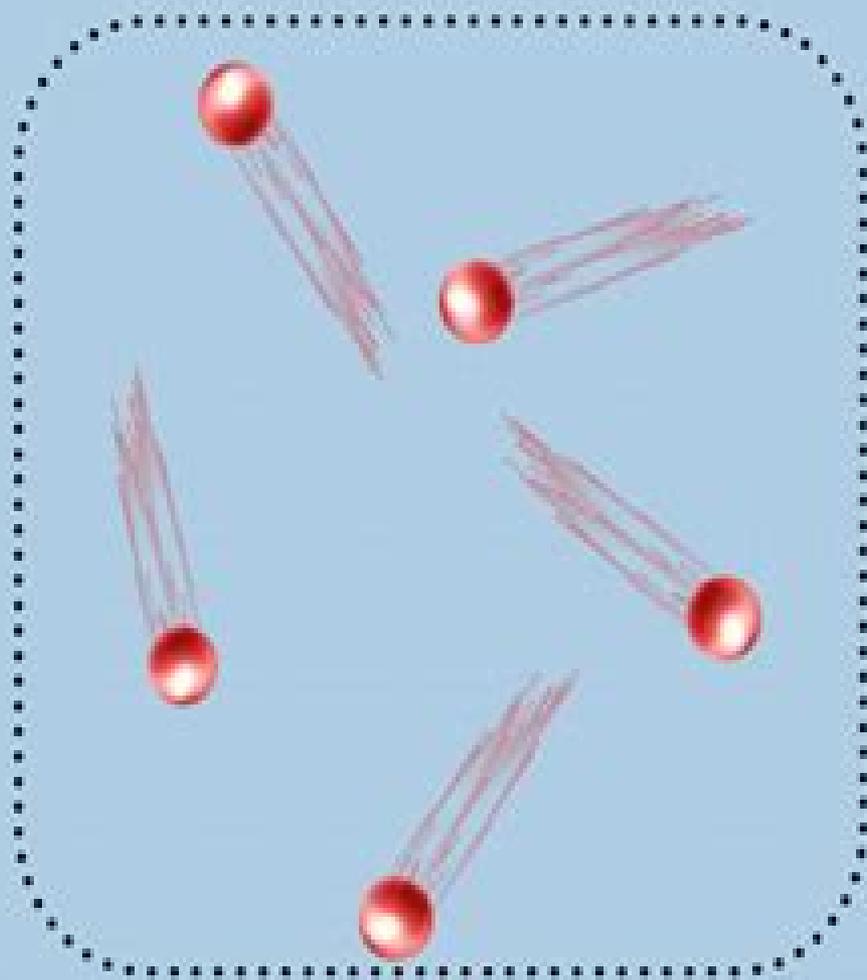


Kinetic Molecular Theory of Gases

- Many gas particles, so statistics applies
- Particles are tiny and spread far apart
- Gas particles are in constant random motion
- No attraction or repulsion between particles
- Collisions are elastic



Kinetic Molecular Theory Instructors Guide

M Lipman



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