

Equilibrium Practice Test

Multiple Choice

Identify the choice that best completes the statement or answers the question.

- _____ 1. A reaction in which products can react to re-form reactants is
a. reforming. c. buffered.
b. reversible. d. impossible.
- _____ 2. At equilibrium,
a. all reactions have ceased.
b. only the forward reaction continues.
c. only the reverse reaction continues.
d. both the forward and reverse reactions continue.
- _____ 3. What symbol in a chemical equation indicates equilibrium?
a. \rightarrow c. \rightleftharpoons
b. \leftarrow d. None of the above
- _____ 4. What two processes are at equilibrium in a saturated sugar solution?
a. evaporation and condensation c. decomposition and synthesis
b. dissolving and crystallization d. ionization and recombination
- _____ 5. What symbol represents the equilibrium constant?
a. k c. c
b. K d. R
- _____ 6. A very high value of K indicates that
a. equilibrium is reached slowly. c. reactants are favored.
b. products are favored. d. equilibrium has been reached.
- _____ 7. The value of K for a system
a. can be calculated from the molar masses of products and reactants.
b. can be calculated from the enthalpies of the forward and reverse reactions.
c. can be calculated from the chemical properties of products and reactants.
d. must be measured by experiment.
- _____ 8. In the equation $K = \frac{[W][X]}{[Y][Z]}$, what represents the concentrations of the reactants?
a. $[Y]$ and $[Z]$ c. $\frac{[W][X]}{[Y][Z]}$
b. $[W]$ and $[X]$ d. $K = \frac{[W][X]}{[Y][Z]}$
- _____ 9. An equilibrium mixture of SO_2 , O_2 , and SO_3 gases at 1500 K is determined to consist of 0.344 mol/L SO_2 , 0.172 mol/L O_2 , and 0.56 mol/L SO_3 . What is the equilibrium constant for the system at this temperature? The balanced equation for this reaction is
$$2\text{SO}_2(\text{g}) + \text{O}_2(\text{g}) \rightleftharpoons 2\text{SO}_3(\text{g})$$

a. 0.41 c. 6.7
b. 2.8 d. 15

Kinetics And Equilibrium Practice Test

Guillaume Favre



Kinetics And Equilibrium Practice Test:

The Enigmatic Realm of **Kinetics And Equilibrium Practice Test**: Unleashing the Language is Inner Magic

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