

Student Exploration: Phase Changes

Approximate time: 45 minutes

Version: 1.0 (2019)

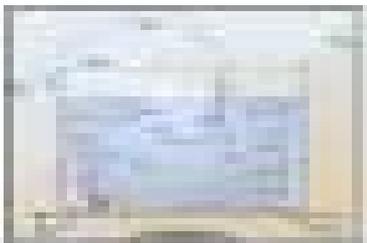
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Prerequisite Knowledge: Students should understand the basic concepts of temperature and heat.

1. In this activity, you will explore the phase changes of water. You will use a virtual simulation to observe the phase changes of water and answer questions about the energy and temperature changes that occur.

Learning Objectives: (The page lists the objectives and the questions.)

Key Concepts: Phase changes, temperature, heat, energy, melting, freezing, boiling, condensation, sublimation, deposition.



1. What happens to the temperature of water as it melts?
 - ANSWER: The temperature remains constant.
2. What happens to the temperature of water as it boils?
 - ANSWER: The temperature remains constant.
3. What happens to the temperature of water as it freezes?
 - ANSWER: The temperature remains constant.
4. What happens to the temperature of water as it condenses?
 - ANSWER: The temperature remains constant.

<p>Key Concepts:</p> <ul style="list-style-type: none"> • Phase changes occur at constant temperature. • Heat is added or removed during phase changes. • The temperature of a substance remains constant during a phase change. 	
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5. **Graphing:** Use the simulation to create a phase change graph. Plot temperature versus time.
 - a. What happens to the temperature of water as it melts?
 - ANSWER: The temperature remains constant.
 - b. What happens to the temperature of water as it boils?
 - ANSWER: The temperature remains constant.
 - c. What happens to the temperature of water as it freezes?
 - ANSWER: The temperature remains constant.
6. **Graphing:** Use the simulation to create a phase change graph. Plot temperature versus time.
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 - c. What happens to the temperature of water as it freezes?
 - ANSWER: The temperature remains constant.



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