

## Ion Worksheet

Symbol	Ion name	Cation or Anion	More or less electrons compared to atom	Number of electrons gained or lost compared to atom
<b>Li<sup>+</sup></b>	<b>Lithium ion</b>	<b>Cation</b>	<b>Less</b>	<b>1</b>
<b>Na<sup>+</sup></b>	Sodium ion	Cation	Less	1
<b>Cl<sup>-</sup></b>	<b>Chloride ion</b>	Anion	More	1
<b>Al<sup>3+</sup></b>	<b>Aluminum ion</b>	Cation	Less	3
<b>S<sup>2-</sup></b>	Sulfide ion	Anion	More	2
<b>Mg<sup>2+</sup></b>	Magnesium ion	Cation	Less	2
<b>Fe<sup>3+</sup></b>	<b>Iron (III) ion</b>	Cation	Less	3
<b>Cu<sup>2+</sup></b>	Copper (II) ion	Cation	Less	2
<b>Cu<sup>+</sup></b>	<b>Copper (I) ion</b>	Cation	Less	1
<b>Fe<sup>2+</sup></b>	Iron (II) ion	Cation	Less	2
<b>O<sup>2-</sup></b>	<b>Oxide ion</b>	Anion	More	2
<b>N<sup>3-</sup></b>	Nitride ion	Anion	More	3
<b>Ca<sup>2+</sup></b>	<b>Calcium ion</b>	Cation	Less	2
<b>I<sup>-</sup></b>	Iodide ion	Anion	More	1
<b>K<sup>+</sup></b>	Potassium ion	Cation	Less	1
<b>Ni<sup>2+</sup></b>	<b>Nickel (II) ion</b>	Cation	Less	2

# Ions Answer Key

**John T. Moore, Richard H. Langley**



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