

Name : _____ Date : _____

Lab Periodic Trends Worksheet

The atomic and the ionic radius of the first 18 elements of the periodic table are given below.

Element	H	He	Li	Be	B	C	N	O	F
Atomic Number	1	2	3	4	5	6	7	8	9
Atomic Radius (Å)	0.79	0.49	2.05	1.4	1.17	0.91	0.75	0.65	0.57
Ionic Radius (Å)	0.012	-	0.76	0.35	0.23	0.16	0.13	1.4	1.33

Ne	Na	Mg	Al	Si	P	S	Cl	Ar
10	11	12	13	14	15	16	17	18
0.38	2.23	1.72	1.82	1.46	1.23	1.09	0.97	0.71
-	1.02	0.72	0.535	0.4	0.38	0.37	1.81	-

Using the data above, construct the line graph of atomic radius vs. atomic number and ionic radius vs. the atomic number.

Atomic Radius vs. Atomic Number



Ionic Radius vs. Atomic Number



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Fields, James M



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different factors in form of correction factors which affect the atomic radii using graphs It takes into accounts the chemical behaviour of all the elements in their respective groups and periods **Explanatory Key to the Periodic Chart of the Atoms** Henry D. Hubbard,1926 *Atomic Radius* , *Calculation of Atomic Radius from Density* Fields, James M,1983

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