

Experiment 1

AIMS:

1. To study the effect of temperature on the rate of polymerization.
2. To study the effect of concentration on the rate of polymerization.
3. To study the effect of catalyst on the rate of polymerization.

APPARATUS:

1. Reaction flask
2. Thermometer
3. Stopwatch
4. Pipette
5. Burette
6. Standard solution
7. Indicator

REAGENTS:

1. Potassium persulphate (K₂S₂O₈)
2. Potassium dichromate (K₂Cr₂O₇)
3. Potassium iodide (KI)

PROCEDURE:

1. Preparation of standard solution
2. Preparation of indicator

Calculation of Rate of Polymerization

Result:

Temp. (°C)	Time (min)	Volume of standard solution (ml)
30	10	10
40	10	10
50	10	10

Conclusion:

Precautions:

Notes:

1. The solution should be prepared in the dark and stored in a dark bottle.
2. The solution should be prepared in the dark and stored in a dark bottle.
3. The solution should be prepared in the dark and stored in a dark bottle.

4. To study the effect of temperature on the rate of polymerization.
5. To study the effect of concentration on the rate of polymerization.
6. To study the effect of catalyst on the rate of polymerization.

7. Reaction flask
8. Thermometer
9. Stopwatch
10. Pipette
11. Burette
12. Standard solution
13. Indicator

14. Potassium persulphate (K₂S₂O₈)
15. Potassium dichromate (K₂Cr₂O₇)
16. Potassium iodide (KI)

17. Preparation of standard solution
18. Preparation of indicator

Temp. (°C)	Time (min)	Volume of standard solution (ml)
30	10	10
40	10	10
50	10	10

19. The solution should be prepared in the dark and stored in a dark bottle.
20. The solution should be prepared in the dark and stored in a dark bottle.



Lab Manual for Students, Department of Polymer Technology

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Laboratory Exercise 52 Part

Lingjun Ying



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