

C4.1 Relative masses and moles

Question number	Answer	Marks	Guidance
1	The mean (average) relative mass of an atom of an element, taking into account the proportions of different isotopes naturally occurring in that element, using the scale carbon-12 = 12.	1 1	
2 a	62	1	
2 b	180	1	
3 a	0.005 moles	1	
3 b i	0.3 moles	1	
3 b ii	500 000 moles	1	
4 a	5000 g or 5 kg	1	
4 b	0.1 g	1	
4 c	74.4 g	1	
5	due to averaging relative masses of different isotopes in a naturally occurring sample	1	

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