

Form 4 Chemistry Notes

Group 1	Group 15	M_3N	Na_3N
Group 1	Group 16	M_2N	K_2O
Group 1	Group 17	MN	$LiCl$
Group 2	Group 15	M_3N_2	Ca_3N_2
Group 2	Group 16	MN	MgO
Group 2	Group 17	MN_2	$BaBr_2$
Group 13	Group 15	MN	AlN
Group 13	Group 16	M_2N_3	Al_2O_3
Group 13	Group 17	MN_3	AlI_3

Covalent Bonding

1. Covalent bonds are formed by atoms sharing electrons to form molecules. This type of bond usually formed between two non-metallic elements.
2. In the formation of covalent bonds, atoms of non-metals will combine with each other to donate one, two or three electrons for sharing.
3. The compound formed through the formation of covalent bonds is called the covalent compounds.

Examples of Covalent Compound

Covalent Compound	Formula	Covalent Compound	Formula
Chlorine	Cl_2	Phosphorus	P_4
Hydrogen	H_2	Sulfur dioxide	SO_2
Oxygen	O_2	Carbon dioxide	CO_2

Form Chemistry Notes

SJ Ball



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