

## Naming and Writing Formulas for Covalent Compounds Worksheet

**Directions:** Write the formula of the following molecular / covalent compounds.

1. Carbon dodecahydride



2. Oxygen trifluoride



3. Heptanitrogen monoxide



4. Tetraphosphorus monobromide



5. Pentasulfur hexachloride



6. Hexaselenium dichloride



7. Selenium decafluoride



8. Trisulfide monofluoride



9. Monocarbon hexanitride



10. Dicarbon diphosphide



11. Carbon tetrafluoride



12. Pentanitrogen hexoxide



**Directions:** Write the name of the following molecular / covalent compounds.

13.  $C_2N_4$

Dicarbon pentanitrogen

14.  $O_2I_2$

Heptaoxygen dichloride

15.  $PO_2$

Phosphorus dioxide

16.  $S_4Cl_2$

Octasulfur trichloride

17.  $Se_4F_2$

Dioxoselenium pentafluoride

18.  $N_4Br_2$

Tetranitrogen tetrabromide

19.  $O_5F_2$

Pentaoxygen pentafluoride

20.  $C_2F_2$

Monocarbon monofluoride

21.  $N_2S_2$

Trinitrogen hexosulfide

22.  $Se_4Br_2$

Troselenium heptabromide

23.  $CS_2$

Carbon disulfide

24.  $P_5$

Phosphorus monosulfide

# Formula Writing For Covalent Compounds

**John W. Hill**



## Formula Writing For Covalent Compounds:

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examining experience Cambridge IGCSE Chemistry Coursebook with CD ROM gives comprehensive and accessible coverage of the syllabus Suggestions for practical activities are included designed to help develop the required experimental skills with full guidance included on the CD ROM Study tips throughout the text exam style questions at the end of each chapter and a host of revision and practice material on the CD ROM are designed to help students prepare for their examinations Answers to the exam style questions in the Coursebook are provided on the CD ROM

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**Rapid Review of Chemistry for the Life Sciences and Engineering** Armen S. Casparian, Gergely Sirokman, Ann Omollo, 2021-12-07 Designed to demystify chemistry for the non chemist Rapid Review of Chemistry for the Life Sciences and Engineering is a useful reference manual for life scientists and engineers who may have forgotten a formula principle or concept in the college chemistry taken a few years ago With over 100 solved examples from balancing chemical reactions doing stoichiometry and understanding nomenclature rules in both organic and inorganic chemistry to calculating half lives in kinetics or radioactive decay schemes understanding colligative properties of solutions and interpreting toxicities of hazardous materials this book is intended to make reviewing and understanding chemistry much clearer and easier Relevant diagrams are in color and solved examples are organized by subject topic and cross referenced by page and chapter number It may also serve as a concise go to sidekick for students who are not chemistry majors taking chemistry at the college level and having difficulty understanding the scope focus language or equations in their chemistry textbook Armed with select contemporary applications it is written in the hope to bridge a gap between chemists and non chemists so that they may communicate with and understand each other Chapters 1-10 are designed to contain the standard material in an introductory college chemistry course Chapters 11-15 present applications of chemistry that should interest and appeal to scientists and engineers engaged in a variety of fields Additional features More than 100 solved examples clearly illustrated and explained with SI units and conversion to other units using conversion tables included Assists the reader to understand organic and inorganic compounds along with their structures including

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