

Nuclear Physics & Nuclear Reactions

Practice Problems

PSI AP Physics B

Name _____

Multiple Choice Questions

- The atomic nucleus consists of:
(A) Electrons
(B) Protons
(C) Protons and electrons
(D) Protons and neutrons
(E) Neutrons and electrons
- The atomic mass number represents the following:
(A) The total number of electrons in the atom
(B) The total number of protons in the atom
(C) The total number of protons and neutrons in the atom
(D) The total number of neutrons in the atom
(E) The total number of electrons and protons in the atom
- The atomic number represents the following:
(A) The number of protons in the atom
(B) The number of neutrons in the atom
(C) The total number of protons and neutrons in the atom
(D) The total number of electrons and neutrons in the atom
(E) The total number of protons and electrons in the atom
- The isotope of carbon $^{14}_6\text{C}$ consists of:

	Protons	Neutrons
(A)	6	6
(B)	6	7
(C)	6	8
(D)	6	14
(E)	6	12
- The isotope of uranium $^{238}_{92}\text{U}$ consists of:

	Protons	Neutrons
(A)	92	143
(B)	92	146
(C)	92	144
(D)	92	145
(E)	92	238

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Huangqi Zhang



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