

Introduction:

The laboratory method used in the experiment is titration. Titration is a method used in measuring the amount of an analytical reagent necessary to react quantitatively with the sample. Acid-base titrations are important for counting concentrations of acids and bases. The point at which the equivalence point has been reached is called the theoretical end point. The titration is detected by some physical change, such as colour change, to indicate the end point. The indicator solution used in this experiment is phenolphthalein which is a colourless in acidic solution. When the reaction reached the endpoint, the solution becomes pale pink. One or two drops of phenolphthalein are sufficient for the titration. Addition of too much indicator will necessitate an indicator blank. The indicator should give a clear visual change in the liquid being titrated once the reaction between the standard solution and the substance is practically complete. The indicators can change colour because their ions have colours that are different from the undissociated molecule (Tan, 2010). The acid and base titrations used the Arrhenius theory. This theory stated that acid are substance which produce hydrogen ions in solution and bases are substance which produce hydroxide ion in solution. In titration, standard solutions are now usually expressed in terms of molarity, M.

Molarity, M = moles of solute/volume of solutions in litres

Objective :

One of the purposes of this experiment is to demonstrate the basic laboratory technique of titration towards the students. This experiment enables a hand-on practice done by the students themselves. Thus, the students will know how to handle the materials, apparatus and the correct way to conduct the experiment as well as being careful to avoid any experimental error. Moreover, this experiment also teach the students on the correct steps to calculate percentage of substance, molarity of solution and other calculations involve throughout the experiment.

Apparatus :

Burette with stand
Pipette
1-L volumetric flask with stopper
250 mL Erlenmeyer flasks
Retort stand with clamp

Reagents :

Sodium hydroxide pellets
Potassium hydrogen phthalate (KHP)
Phenolphthalein
Unknown vinegar

Introduction In A Lab Report

**Andrea A. Gilpin, Patricia Patchet-
Golubev**



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A Guide to Writing in the Sciences Andrea A. Gilpin, Patricia Patchet-Golubev, 2000-01-01 Clear and concise this guide describes the basic elements of scientific writing from lab reports to research essays to articles as well as the grammar and punctuation fundamental to all writing 128 pp

Fluid Mechanics Experiments Robabeh Jazaei, 2022-05-31 Fluid mechanics is one of the most challenging undergraduate courses for engineering students The fluid mechanics lab facilitates students learning in a hands on environment The primary objective of this book is to provide a graphical lab manual for the fluid mechanics laboratory The manual is divided into six chapters to cover the main topics of undergraduate level fluid mechanics Chapter 1 begins with an overview of laboratory objectives and the introduction of technical laboratory report content In Chapter 1 error analysis is discussed by providing examples In Chapter 2 fluid properties including viscosity density temperature specific weight and specific gravity are discussed Chapter 3 revolves around the fluid statics include pressure measurement using piezometers and manometers Additionally hydrostatic pressure on the submerged plane and curved surfaces as well as buoyancy and Archimedes Principle are examined in Chapter 3 In Chapter 4 several core concepts of fluid dynamics are discussed This chapter begins with defining a control system based on which momentum analysis of the flow system is explained The rest of the chapter is allotted to the force acting on a control system the linear momentum equation and the energy equation Chapter 4 also covers the hydraulic grade line and energy grade line experiment The effect of orifice and changing cross sectional area by using Bernoulli s equation is presented in Chapter 4 The application of the siphon is extended from Chapter 4 by applying Bernoulli s equation The last two chapters cover various topics in both internal and external flows which are of great importance in engineering design Chapter 5 deals with internal flow including Reynolds number flow classification flow rate measurement and velocity profile The last experiment in Chapter 5 is devoted to a deep understanding of internal flow concepts in a piping system In this experiment students learn how to measure minor and major head losses as well as the impact of piping materials on the hydrodynamics behavior of the flow Finally open channels weirs specific energy and flow classification hydraulic jump and sluice gate experiments are covered in Chapter 6

Laboratory Manual, Direct and Alternating Current Clarence Edward Clewell, 1924

Intelligent Tutoring Systems Scott Crossley, Elvira Popescu, 2022-06-24 This volume constitutes the proceedings of the 18th International Conference on Intelligent Tutoring Systems ITS 2022 held in Bucharest Romania in June 2022 The 14 full papers 13 short papers and 11 poster papers presented in this volume were carefully reviewed and selected from 50 submissions The papers are categorized into the following topical sub headings Tools and Methods for learning Sciences and Practices Algorithms for Prediction Recommendation and Classification in Learning Systems Tutoring and Learning Systems New Approaches Framework and Theories

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Laboratory Investigations in Anatomy and Physiology Stephen N. Sarikas, 2007 This concise lab manual is designed for instructors who wish to avoid cookbook style lab instruction for Anatomy Physiology Through the use of an engaging connective learning methodology author Stephen Sarikas builds each lab exercise step on the previous one helping readers to understand complex ideas and make connections between concepts

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KEY MARKET For all readers interested in anatomy physiology labs

The Fundamentals of Scientific Research Marcy A. Kelly, 2015-08-06 The Fundamentals of Scientific Research An Introductory Laboratory Manual is a laboratory manual geared towards first semester undergraduates enrolled in general biology courses focusing on cell biology This laboratory curriculum centers on studying a single organism throughout the entire semester *Serratia marcescens* or *S. marcescens* a bacterium unique in its production of the red pigment prodigiosin The manual separates the laboratory course into two separate modules The first module familiarizes students with the organism and lab

equipment by performing growth curves Lowry protein assays quantifying prodigiosin and ATP production and by performing complementation studies to understand the biochemical pathway responsible for prodigiosin production Students learn to use Microsoft Excel to prepare and present data in graphical format and how to calculate their data into meaningful numbers that can be compared across experiments The second module requires that the students employ UV mutagenesis to generate hyper pigmented mutants of *S marcescens* for further characterization Students use experimental data and protocols learned in the first module to help them develop their own hypotheses experimental protocols and to analyze their own data Before each lab students are required to answer questions designed to probe their understanding of required pre laboratory reading materials Questions also guide the students through the development of hypotheses and predictions Following each laboratory students then answer a series of post laboratory questions to guide them through the presentation and analysis of their data and how to place their data into the context of primary literature Students are also asked to review their initial hypotheses and predictions to determine if their conclusions are supportive A formal laboratory report is also to be completed after each module in a format similar to that of primary scientific literature The Fundamentals of Scientific Research An Introductory Laboratory Manual is an invaluable resource to undergraduates majoring in the life sciences

Laboratory Manual Clarence Edward Clewell,1914 *Methods in Biotechnology* Seung-Beom Hong,M. Bazlur Rashid,Lory Z. Santiago-Vázquez,2016-08-01 As rapid advances in biotechnology occur there is a need for a pedagogical tool to aid current students and laboratory professionals in biotechnological methods *Methods in Biotechnology* is an invaluable resource for those students and professionals *Methods in Biotechnology* engages the reader by implementing an active learning approach provided advanced study questions as well as pre and post lab questions for each lab protocol These self directed study sections encourage the reader to not just perform experiments but to engage with the material on a higher level utilizing critical thinking and troubleshooting skills This text is broken into three sections based on level *Methods in Biotechnology* *Advanced Methods in Biotechnology I* and *Advanced Methods in Biotechnology II* Each section contains 14 22 lab exercises with instructor notes in appendices as well as an answer guide as a part of the book companion site This text will be an excellent resource for both students and laboratory professionals in the biotechnology field [Report of the President of Harvard College and Reports of Departments](#) Harvard University,1921 **Annual Report** New York State Agricultural Experiment Station,1895 **Biennial Report of the President of the University on Behalf of the Board of Regents** University of California (System),1914 [A Laboratory Manual for the Study of General Botany](#) William Evans Lawrence,1930 *Annual Report of the President of the University on Behalf of the Regents to His Excellency the Governor of the State of California* University of California, Berkeley,1912 **Annual Report of the President of the University on Behalf of the Regents to His Excellency the Governor of the State of California** University of California (1868-1952). President,1915 [PEOPLE OF THE STATE OF MICHIGAN V MERL JONATHAN FAUCETT, 442 MICH 153 \(1993\)](#) ,1993

93765 *Laboratory Manual for Fundamental Chemistry* Donald Hatch Andrews, Richard J. Kokes, 1962 *Annual Report of the Board of Control of the New York Agricultural Experiment Station, (Geneva, Ontario County), ... , with Reports of Director and Other Officers* New York State Agricultural Experiment Station, 1893 Technical Report Great Britain. Advisory Committee for Aeronautics, 1914 **Technical Report** Aeronautical Research Council (Great Britain), 1914
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