

**ICSE Board
Class X Chemistry
Board Paper - 2009**

Time: 1½ hrs

Total Marks: 80

General Instructions:

1. Answers to this paper must be written on the paper provided separately.
2. You will NOT be allowed to write during the first 15 minutes. This time is to be spent in reading the question paper.
The time given at the head of this paper is the time allowed for writing the answers.
This question paper is divided into two sections.
3. Section I contains one question with parts (a) to (h); all the eight parts are to be answered.
4. Section II contains six questions numbered 2 to 7. You are to answer any four of these questions.
The intended marks of questions or for parts of questions are given in brackets [].

SECTION I (40 Marks)

Attempt all questions from this section.

Question 1

- (a)** Name the gas evolved in each case (formula is not acceptable). [5]
- i. The gas produced by the action of concentrated sulphuric acid on sodium chloride.
 - ii. The gas produced by the action of dilute nitric acid on copper.
 - iii. The gas produced on heating sodium nitrate.
 - iv. The gas which burns in oxygen with a green flame.
 - v. The gas which can be oxidised to sulphur. **
- (b)** Match each substance A to E listed below with the appropriate description in parts (i) to (v). [5]
- (A) Sulphur (B) Silver chloride (C) Hydrogen chloride (D) Copper (II) sulphate
(E) Graphite
- i. A non-metal which is a good conductor of electricity.
 - ii. A covalent compound which behaves like an ionic compound in aqueous solution.
 - iii. A compound which is insoluble in cold water but soluble in excess of ammonia solution.
 - iv. A pink metal which is deposited at the cathode during the electrolysis of the solution of this salt.
 - v. A non-metal which reacts with concentrated nitric acid to form its own acid as one of the product.

Ksscience Paper 2009

Liyong Dong



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