

Question	Answer	Mark	Guidance															
2 (a)	<p>FIRST CHECK THE ANSWER ON THE ANSWER LINE IF answer = $\text{CH}_4\text{N}_2\text{O}$ award 2 marks</p> <table border="0" style="width: 100%; text-align: center;"> <tr> <td>C</td> <td>H</td> <td>N</td> <td>O</td> <td></td> </tr> <tr> <td>20.00/12.0</td> <td>6.67/1.0</td> <td>46.67/14.0</td> <td>26.66/16.0</td> <td></td> </tr> </table> <p>OR</p> <table border="0" style="width: 100%; text-align: center;"> <tr> <td>1.67</td> <td>6.67</td> <td>3.33</td> <td>1.67</td> <td>ratio of mol ✓</td> </tr> </table> <p>to give $\text{CH}_4\text{N}_2\text{O}$ ✓</p>	C	H	N	O		20.00/12.0	6.67/1.0	46.67/14.0	26.66/16.0		1.67	6.67	3.33	1.67	ratio of mol ✓	2	<p>ALLOW 1.66 for C OR 1.66 for O</p> <p>IGNORE Significant figures beyond the 3rd significant figure. (eg ALLOW 3.3335 for N OR 1.666 for C)</p> <p>ALLOW ECF from incorrectly calculated ratio of mol. DO NOT ALLOW ECF from using an atomic number OR any original sums inverted (eg 12.00/20.00) ALLOW any order of atoms</p>
C	H	N	O															
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2 (c) (i)	H_3PO_4 ✓	1	ALLOW formula if seen as reactant in an equation IGNORE name															
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	Total	6																

F321 Ocr 2014

Marco Casella



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