

11. Flow rate of granules from the hopper can be improved by adding  
(A) Lubricant (B) Disintegrant (C) Binder (D) Glidant
12. Sterilization temperature for aqueous solution in autoclave is  
(A) 60°C (B) 121°C (C) 160°C (D) 180°C
13. The vitamin administered with isoniazid to minimize its adverse reaction is  
(A) Vitamin A (B) Pyridoxine (C) Biotin (D) Pantothenic acid
14. State Pharmacy council should have the following number of elected members  
(A) Six (B) Nine (C) Five (D) Seven
15. Synthesis of Urea takes place exclusively  
(A) Kidney (B) Liver (C) Gall bladder (D) Urinary bladder
16. How many parts of 10% alcohol be mixed with 2 parts of 15% alcohol to get 12% alcohol ?  
(A) 2 (B) 3 (C) 5 (D) 6
17. The UV- Visible region in the electromagnetic spectrum of radiation is  
(A) 200-400 nm (B) 300-600 nm (C) 400-800 nm (D) 200-800 nm
18. In Drugs & Cosmetics Act and Rules there under, list of substances that should be retail on prescription of Registered Medical Practitioner is given in  
(A) Schedule H (B) Schedule G (C) Schedule J (D) Schedule O
19. Alkaloids in cinchona bark are detected by  
(A) Iodine test (B) Thalleioquin test  
(C) Liebermann-Burchard test (D) Nessler's test

# Lateral Entry Question Papers

**Alan Grant, Law Reform Commission of  
Canada**

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**M.Com Entrance Question Paper ( Subject wise bifurcated) - Delhi University** Dheeraj Kumar Singh, This book contains previous year question asked in M Com Entrance Delhi University All Question Paper is divided in 4 groups and then Subject wise 1 Financial Accounting 5 19 2 Corporate Accounting 20 30 3 Cost Management Accounting 31 52 4 Business Mathematics 53 71 5 Business Statistics 72 94 6 Macro Economics Indian Economy 95 117 7 Micro Economics 118 137 8 Income Tax 138 150 9 Business Law 151 162 10 Business Management 163 175 11 Corporate Law 176 185 Financial Accounting 2020 7 Questions 1 In case of rising prices inflation FIFO method will a Provide lowest value of closing stock and profit b Provide highest value of closing stock and profit c Provide highest value of closing stock but lowest value of profit d Provide highest value of profit but lowest value of closing stock 2 Which of the following will cause difference on a trial balance a An invoice omitted from the sale day book b An invoice of Rs 1 500 entered in the sale day book as Rs 15 000 c An invoice of Rs 5 000 entered in the sale day book not included in the monthly total d A credit note entered to the sales day book 3 International Accounting Standards IAS have been developed by a The European commission b The International accounting standard committee c The financial reporting council d All of these 4 Which one of the following item cannot be recorded in the appropriation account in case of partnership accounting a Interest on capital b Interest on drawings c Rent paid to partners d Partners salary 5 In the absence of partnership deed interest on partners loan is payable by firm a 12% p a b 6% p a c 10% p a d 16% p a 6 Royalties revenue should be recognized a On an accrual basis in accordance with the terms of the relevant agreement b On cash basis c On an actual basis d Revenue is recognized on the time proportion basis 7 AS 29 is applicable only to a Provision for valuation adjustment for fixed Assets b Provision for Liabilities and charges c Provision for valuation adjustment for current assets d Furniture

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web elements in the second row of the periodic table place their electrons in the  $2n$  shell as well as the  $1n$  shell for instance lithium li has three electrons two fill the  $1s$  orbital and the third is placed in the  $2s$  orbital giving an electron configuration of  $1s^2 2s^1$

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web electron configurations and trends answer these discussion questions given the following part of an electron configuration  $3s^2$  explain which part refers to the number of electrons the energy level and the sublevel what is the maximum number of electrons that can occupy a single orbital state and explain hund's rule

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web one more point needs to be emphasized about the relationship between electron configuration and the periodic table the atoms of elements in the same vertical column of the table have similar electron configurations for example consider the alkaline earth elements group iia

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web the electron removed during the ionization of beryllium  $he 2s^2$  is an  $s$  electron whereas the electron removed during the ionization of boron  $he 2s^2 2p^1$  is a  $p$  electron this results in a lower first ionization energy for boron even though its nuclear charge is greater by one proton

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web predictions from reliable sources have been used for these elements grayed out electron numbers indicate subshells filled to their maximum bracketed noble gas symbols on the left represent inner configurations that are the same in each period written out these are  $he 2$  helium  $1s^2$   $ne 10$  neon  $1s^2 2s^2 2p^6$

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web q1 what is the electron configuration of the element potassium answer the atomic number of potassium is 19 a neutral potassium element has 19 electrons its electron configuration is  $1s^2 2s^2 2p^6 3s^2 3p^6 4s^1$  or simply put as  $4s^1$  q2 what is the total number of electrons that can be filled in  $s$   $p$  and  $d$  subshells

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web by building up from hydrogen this table can be used to determine the electron configuration for any atom on the periodic table we will now construct the ground state electron configuration and orbital diagram for a selection of atoms in the first and second periods of the periodic table

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web electron configurations are a simple way of writing down the locations of all of the electrons in an atom as we know the positively charged protons in the nucleus of an atom tend to attract negatively charged electrons

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