

Exam 3 Fundamentals Study Guide

Anatomy and Physiology:

- **Kidneys:** Assist with RBC production and BP regulation.
- **Fallopian Tubes:** Carry ova from ovaries to the uterus.
- **Cardiac Landmarks:**
 - **Aortic:** Right of sternum/2nd ICS
 - **Pulmonic:** Left of sternum/2nd ICS
 - **Erb's Point:** Left of sternum/3rd ICS
 - **Tricuspid:** Left of sternum/4th ICS
 - **Mitral:** Left of sternum/5th ICS
- **Prostate Gland:** Located surrounding the urethra near the bladder.
- **Liver Span:** Normal size range is 2.5-3 inches.

Clinical Skills and Assessment:

- **Blumberg Sign:** Assessing for rebound tenderness in RLQ for appendicitis.
- **Assessing Pulses:** Use Doppler ultrasound for non-palpable pulses in patients with bilateral lower extremity edema.
- **Cardiac Assessment:** Auscultate heart sounds, palpate pulses, and check capillary refill. Do not percuss pericardium.
- **Rectal Exam Positioning:** The patient should be placed in a side-lying position.

Pathophysiology and Disease Processes:

- **Hypovolemia Risk Factors:** Excessive nausea, vomiting, diarrhea.
- **Venous Insufficiency:** Results in bilateral edema of the lower extremity.
- **Lymphedema:** Results in unilateral edema of the lower extremity.
- **DVT (Deep Vein Thrombosis):** Characterized by pain, redness, swelling, and a positive Homan sign in the calf.
- **UTIs in Geriatrics:** The first sign is often acute mental confusion.
- **Viral Hepatitis Causes:** Contaminated stool/food/water, undercooked/raw shellfish, sexual contact, sharing needles, close contact with infected person.
- **Bowel Obstruction:** NPO to allow bowel rest is an appropriate intervention.

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LP Steffe



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web in the kitchen for nancy one night we waked up hearing the sound it was not singing and it was not crying coming up the dark stairs there was a light in mother s room and we heard father going down the hall down the back stairs and caddy and i went into the hall the floor was cold

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web that evening sun quentin compson reminisces about his hometown of jefferson mississippi which has changed a great deal since he was a child the town has been modernized with paved streets telephone poles and a city laundry even the negro women now have cars which they use when they are doing the laundry for the white

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web the lyrics of this song refer to a woman who hates to see that evening sun go down because it makes her feel that she is going to die it also refers to the fact that the woman is going to make her getaway if she still feels afraid in the morning

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web i hate to see that evening sun go down i hate to see that evening sun go down cause my baby he s gone left this town feelin tomorrow like i feel today if i m feelin tomorrow like i feel today i ll pack my truck and make my give a wayst

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web i can t swallow it nancy said i swallows but it won t go down me you go down to the cabin dilsey said frony will fix you a pallet and i ll be there soon

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web william faulkner that evening sun monday is no different from any other weekday in jefferson now the streets are paved now and the telephone and electric companies are cutting down more and more of the shade trees the water oaks the maples and locusts and elms to make room for iron poles bearing clusters of bloated and ghostly

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web that evening sun first appeared in the march 1931 issue of american mercury the remainder of its publishing history is identical to a rose for emily reprinted in these thirteen 1931 in faulkner s collected stories 1950 and in the selected short stories of william faulkner 1961

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 $2 \text{ CO}_2(\text{g}) + \text{H}_2\text{O}(\text{g}) + \text{C}_2\text{H}_2(\text{g}) \rightarrow \text{C}_2\text{H}_4(\text{g}) + 2 \text{ H}_2\text{O}(\text{g}) + \text{C}_2\text{H}_6(\text{g})$ $\Delta h = 94.5 \text{ kJ}$
 $\text{H}_2(\text{g}) + \frac{1}{2} \text{O}_2(\text{g}) \rightarrow \text{H}_2\text{O}(\text{g})$ $\Delta h = 71.2 \text{ kJ}$
 $\text{C}_2\text{H}_6(\text{g}) + \frac{7}{2} \text{O}_2(\text{g}) \rightarrow 2 \text{ CO}_2(\text{g}) + 3 \text{ H}_2\text{O}(\text{g})$ $\Delta h = 283 \text{ kJ}$

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web chemistry 120 hess s law worksheet calculate Δh for the reaction $\text{C}_2\text{H}_4(\text{g}) + \text{H}_2(\text{g}) \rightarrow \text{C}_2\text{H}_6(\text{g})$ from the following data
 $\text{C}_2\text{H}_4(\text{g}) + 3 \text{ O}_2(\text{g}) \rightarrow 2 \text{ CO}_2(\text{g}) + 2 \text{ H}_2\text{O}(\text{l})$ $\Delta h = 1411 \text{ kJ}$ mole
 $\text{C}_2\text{H}_6(\text{g}) + 7 \text{ O}_2(\text{g}) \rightarrow 2 \text{ CO}_2(\text{g}) + 3 \text{ H}_2\text{O}(\text{l})$ $\Delta h = 1560 \text{ kJ}$ mole
 $\text{H}_2(\text{g}) + \frac{1}{2} \text{O}_2(\text{g}) \rightarrow \text{H}_2\text{O}(\text{l})$ $\Delta h = 285 \text{ kJ}$ mole

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 hess s law example 12.1 determine Δh for the target reaction $\text{C}_2\text{H}_4(\text{g}) + \text{H}_2(\text{g}) \rightarrow \text{C}_2\text{H}_6(\text{g})$ given the following information

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 libretexts quote hess s law quote 2019 which determines the enthalpy change value
 hess s law is written as $\Delta h_{\text{target}} = \Delta h_1 + \Delta h_2 + \Delta h_3$ or $\Delta h_{\text{target}} = \sum \Delta h_{\text{known}}$

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web hess s law states that if a process can be expressed as the sum of two or more steps the enthalpy change for the overall process is the sum of the Δh values for each step to use hess s law two principles must be understood one if an equation is reversed the sign of the Δh value is also reversed two if an equation is multiplied by a

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web hess s law states that the total enthalpy change for the reaction will be the sum of all those changes no matter how many different steps or stages in the reaction there are
 cohen 2016 the equations for the reactions in the experiment done are as follows
 $\text{NaOH}(\text{s}) + \text{Na}(\text{aq}) \rightarrow \text{OH}(\text{aq})$

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web this activity provides a demonstration of hess law using three reactions the solubility naoh in water the solubility naoh in hcl and the reaction of a solution of hcl and a solution of naoh

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web mar 13 2023 be able to use hess s law to calculate reaction enthalpies first law of thermodynamics energy is the ability to do work or transfer heat work is the transfer of energy from one body to another in a sense work is energy in the process of transfer

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web may 8 2015 hess law lab by maya parks partners ben seufert kelsea floyd convert joules to kj in your final answer multiply the mass by the change in temperature and the c given trial 1 use hess s law and the following equations and Δh values to determine the heat of reaction for the reaction $c_2h_4(g)$

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web thus for reaction 9 2 in which 1 mole of h_3o^+ and 1 mole of oh^- combine to form 2 moles of h_2o $\Delta h = 55.8 \text{ kJ}$ as shown note that the enthalpy change $\Delta h_{products} - \Delta h_{reactants}$ is positive if heat is absorbed that is if $\Delta h_{products}$

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web objective the purpose of the lab is to combine two reactants in the calorimeter and measure the heats of reaction in solution eventually allowing an individual to prove or disprove hess s law introduction thermodynamics is the study of energy exchanged between the system and the surroundings

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web 3 12 hess s law in this experiment you will measure the amount of heat released in these three related exothermic reactions 1 naoh s na aq oh aq h 1 2 naoh s h aq cl aq h 2 o na aq cl aq h 2 3

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