

GAS LAWS

INTERACTIVE NOTEBOOK

FOLDABLES

KEY

$P_1 V_1 = P_2 V_2$	GAS LAWS QUANTITIES AND CONSTANTS P = PRESSURE V = VOLUME T = TEMPERATURE n = NUMBER OF MOLES R = UNIVERSAL GAS CONSTANT	$\frac{P_1 V_1}{T_1} = \frac{P_2 V_2}{T_2}$
$\frac{P_1}{P_2} = \frac{V_2}{V_1}$		$\frac{V_1}{T_1} = \frac{V_2}{T_2}$
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 Behind flaps, write the equation for each gas law.

GAS LAWS

BOYLE'S LAW	GAS LAWS QUANTITIES AND CONSTANTS P = V = T = n = R =	COMBINED GAS LAW
GAY-LUSSAC'S LAW		AVOGADRO'S LAW
CHARLES'S LAW		IDEAL GAS LAW

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