

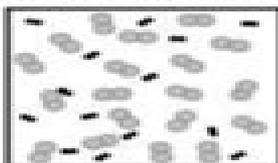
## CHAPTER 18 HOW CHEMICALS MIX

### Multiple Choice

Identify the choice that best completes the statement or answers the question.

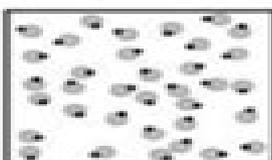
- \_\_\_\_\_ 1. A combination of two or more substances in which they no longer retain their chemical properties is called a(n) \_\_\_\_\_.
- a. heterogeneous mixture
  - b. compound
  - c. suspension
  - d. mixture
  - e. periodic trend

- \_\_\_\_\_ 2. The following image represents which kind of matter?



- a. a mixture
- b. an element
- c. a compound
- d. none of the above
- e. all of the above

- \_\_\_\_\_ 3. The following image represents which kind of matter?



- a. an element
- b. a compound
- c. a mixture
- d. none of the above
- e. all of the above

- \_\_\_\_\_ 4. Which of the following is a mixture?

- a. salt
- b. iron
- c. gold
- d. helium
- e. air

- \_\_\_\_\_ 5. How does a suspension differ from a solution?

- a. A suspension is a heterogeneous mixture consisting of different phases whereas a solution is a homogeneous mixture consisting of a single phase.
- b. The difference between a suspension and a solution can only be determined by chemical means.
- c. Although a solution and suspension are both homogeneous mixtures, only the components of a suspension will separate by spinning the mixture in a centrifuge.
- d. A suspension is a heterogeneous mixture whose components can be separated by simple filtration. A solution is a homogeneous mixture which cannot be separated by simple

# How Chemicals Mix Guide Answers

**Richard Bailey**



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