

**Ionic Compounds: Names and Formulas Worksheet**

1. Write the formulas for the following compounds

Name	Formula	Name	Formula
magnesium oxide	MgO	copper (I) bromide	CuBr
sodium fluoride	NaF	tin (II) iodide	SnI <sub>2</sub>
aluminum nitride	AlN	iron (III) chloride	FeCl <sub>3</sub>
potassium sulfide	K <sub>2</sub> S	calcium phosphide	Ca <sub>3</sub> P <sub>2</sub>
lithium iodide	LiI	lead (II) oxide	PbO
calcium bromide	CaBr <sub>2</sub>	lead (IV) fluoride	PbF <sub>4</sub>
beryllium oxide	BeO	tin (IV) bromide	SnBr <sub>4</sub>
nickel (II) chloride	NiCl <sub>2</sub>	copper (II) sulfide	CuS
magnesium nitride	Mg <sub>3</sub> N <sub>2</sub>	iron (II) oxide	FeO
aluminum sulfide	Al <sub>2</sub> S <sub>3</sub>	calcium nitride	Ca <sub>3</sub> N <sub>2</sub>

2. Write the names for the following compounds.

Formula	Name	Formula	Name
Li <sub>2</sub> O	lithium oxide	PbS	lead (II) sulfide
AlCl <sub>3</sub>	aluminum chloride	SnO <sub>2</sub>	tin (IV) oxide
MgS	magnesium sulfide	Na <sub>2</sub> S	sodium sulfide
CaO	calcium oxide	Mg <sub>3</sub> P <sub>2</sub>	magnesium phosphide
KBr	potassium bromide	FeO	iron (II) oxide
BeF <sub>2</sub>	beryllium fluoride	CuI	copper (I) iodide
Na <sub>3</sub> N	sodium nitride	PbCl <sub>4</sub>	lead (IV) chloride
Al <sub>2</sub> O <sub>3</sub>	aluminum oxide	FeP	iron (III) phosphide
CuCl <sub>2</sub>	copper (II) chloride	CaF <sub>2</sub>	calcium fluoride
BeBr <sub>2</sub>	beryllium bromide	K <sub>3</sub> P	potassium phosphide

# Names And Formulas Worksheet Answers

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Staff, Macmillan Publishing, MacMillan  
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web this situation creates large obstacles for the investigation of computer systems and files which are the subject of forensic events especially if a suspect uses a computer system with encrypted files to obtain evidence firstly it is

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