

Gas Stoichiometry #2

How many liters of ammonia can be produced when 12 liters of hydrogen react with an excess of nitrogen?



$$\frac{12 \text{ L } \cancel{\text{H}_2}}{\underline{3} \text{ L } \cancel{\text{H}_2}} \times \frac{2 \text{ L NH}_3}{\underline{2} \text{ L NH}_3} = 8.0 \text{ L NH}_3$$

Gas Stoichiometry Edgenuity

J Elliott

A decorative graphic element consisting of a horizontal light blue bar with a rounded right end, overlaid on a red-to-white gradient semi-circle.

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