

Atomic Structure



Atomic Number

Element Symbol

Element Name

Atomic Mass

The atomic number represents these two parts of the atom:

number of protons & number of electrons

You can calculate the number of neutrons by subtracting

Atomic Mass minus Atomic Number



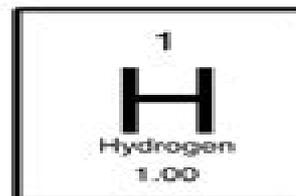
Protons:

7

Electrons:

7

Neutrons:

7

Protons:

1

Electrons:

1

Neutrons:

0

Protons:

8

Electrons:

8

Neutrons:

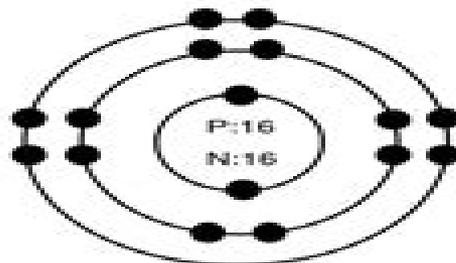
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BOHR DIAGRAMS

- Draw a circle in the center representing the nucleus.
- Label the number of protons and neutrons.
- Draw the electrons in their energy levels.
 1. The first energy level can hold 2 electrons
 2. The second energy level can hold 8 electrons
 3. The third energy level can hold 8 electrons

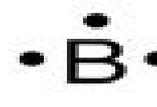
The electrons in the outermost energy level are called:

Valence Electrons

Element: SulfurNumber of Valence Electrons: 6

LEWIS DIAGRAMS

A Lewis diagram shows only the symbol with the Valence electrons



Matching Worksheet For Atomic Structure Answer Key

JG Myers



Matching Worksheet For Atomic Structure Answer Key:

Reviewing **Matching Worksheet For Atomic Structure Answer Key**: Unlocking the Spellbinding Force of Linguistics

In a fast-paced world fueled by information and interconnectivity, the spellbinding force of linguistics has acquired newfound prominence. Its capacity to evoke emotions, stimulate contemplation, and stimulate metamorphosis is truly astonishing. Within the pages of "**Matching Worksheet For Atomic Structure Answer Key**," an enthralling opus penned by a very acclaimed wordsmith, readers attempt an immersive expedition to unravel the intricate significance of language and its indelible imprint on our lives. Throughout this assessment, we shall delve into the book's central motifs, appraise its distinctive narrative style, and gauge its overarching influence on the minds of its readers.

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