

Heat Study Guide

Section 1: States of Matter

1.

State of Matter	Volume (definite or indefinite)	Shape (definite or indefinite)	How do the particles move? Low energy
Solid	definite	definite	Vibrate in place
Liquid	definite	indefinite	Slide past each other
Gas	indefinite	indefinite	Random High energy

Section 2: Change of state – endothermic and exothermic changes

2. What does the term "change of state" mean?

a substance moves from one state of matter to another

3. What causes a change in state?

a change in the energy

4. Fill in the table below.

Condensation	Gas to liquid
freezing	Liquid to solid
Vaporization	Liquid to gas
Sublimation	Solid to gas
Melting	Solid to liquid

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