

Le Chatelier's Principle Practice Worksheet

Complete the following chart by writing left, right, or none for equilibrium shift and decreases, increases, or remains the same for reactants' and products' concentration and for the value of K.

(Remember pure solids and liquids do not affect equilibrium values)



Stress	Equilibrium Shift	[H ₂]	[I ₂]	[HI]	[K]
1. Add H ₂					
2. Add I ₂					
3. Add HI					
4. Remove H ₂					
5. Remove I ₂					
6. Remove HI					
7. Increase Temperature					
8. Decrease Temperature					
9. Increase Pressure					
10. Decrease Pressure					

Le Chateliers Principle Answer Key

Thomas Griffiths



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