

# CLASSIFYING MATTER

## READING COMPREHENSION WORKSHEET

<p>Complete each sentence by filling in the missing information.</p> <p>1. A pure substance made of only one type of atom is called _____.</p> <p>2. How is matter classified? _____.</p> <p>3. The number of protons in the nucleus of an atom is called its _____.</p> <p>4. A neutron has the same mass as a proton but no _____.</p> <p>5. The number of electrons in an atom is equal to the number of _____.</p> <p>Draw a line from each element to its correct family.</p> <p>6. Helium _____ noble gases</p> <p>7. Sodium _____ metals</p> <p>8. Chlorine _____ nonmetals</p> <p>9. Oxygen _____ gases</p> <p>10. Mercury _____ liquids</p> <p>11. Bromine _____ metalloids</p> <p>Answer in 2-3 complete sentences.</p> <p>12. Why is hydrogen a different element than helium?</p> <p>_____</p> <p>13. How does chlorine differ from sulfur in terms of bonding?</p> <p>_____</p> <p>14. How is a carbon atom different from a nitrogen atom?</p> <p>_____</p> <p>15. How does iron differ from copper in terms of conductivity?</p> <p>_____</p>	<p>Read each statement and decide if it is True or False.</p> <p>16. Elements are made of only one kind of atom.</p> <p>17. Matter is made of molecules and atoms.</p> <p>18. Atoms are the smallest particles of matter.</p> <p>19. Protons and neutrons have equal masses.</p> <p>20. Electrons are the lightest particles in an atom.</p> <p>21. Helium is a noble gas.</p> <p>22. Chlorine is a nonmetal.</p> <p>23. Oxygen is a gas.</p> <p>24. Mercury is a liquid.</p> <p>25. Bromine is a metalloid.</p> <p>26. Hydrogen is a different element than helium.</p> <p>27. Chlorine differs from sulfur in terms of bonding.</p> <p>28. Carbon is different from nitrogen in terms of mass.</p> <p>29. Iron and copper differ in terms of conductivity.</p>	<p>1. Solids</p> <p>2. Soluble materials within a homogeneous matter</p> <p>3. Heterogeneous mixture</p> <p>4. Homogeneous mixture</p> <p>5. Heterogeneous mixture</p> <p>6. Homogeneous mixture</p> <p>7. Heterogeneous mixture</p> <p>8. What is the difference between a homogeneous and a heterogeneous mixture?</p> <p>9. They have uniform composition.</p> <p>10. They do not have uniform composition.</p> <p>11. They contain different substances.</p> <p>12. They have uniform composition.</p> <p>13. The number of particles in the large homogeneous mixture</p> <p>14. None</p> <p>15. Heterogeneous mixture</p> <p>16. Homogeneous mixture</p> <p>17. None</p> <p>18. What are the following elements?</p> <p>19. Helium</p> <p>20. Chlorine</p> <p>21. Oxygen</p> <p>22. Mercury</p> <p>23. Bromine</p> <p>24. Hydrogen</p> <p>25. Soluble materials within a homogeneous mixture</p> <p>26. Heterogeneous mixture</p> <p>27. None</p> <p>28. Homogeneous mixture</p> <p>29. Heterogeneous mixture</p> <p>30. Heterogeneous mixture</p> <p>31. Homogeneous mixture</p> <p>32. Heterogeneous mixture</p> <p>33. None</p>	<p style="text-align: center;"><b>Classifying Matter</b></p> <hr/> <p>Read the passage carefully before answering the questions that follow.</p> <div style="border: 1px solid black; padding: 5px;"> <p><b>Working Passage</b></p> <p>Some matter is different from other matter. It is made of particles and molecules. It has different properties and characteristics. Some matter is made of only one type of atom and is called an element. Some matter is made of two or more different types of atoms and is called a compound. Some matter is made of two or more different types of atoms and is called a mixture. Some matter is made of two or more different types of atoms and is called a solution. Some matter is made of two or more different types of atoms and is called a suspension. Some matter is made of two or more different types of atoms and is called a colloid. Some matter is made of two or more different types of atoms and is called a plasma. Some matter is made of two or more different types of atoms and is called a solid. Some matter is made of two or more different types of atoms and is called a liquid. Some matter is made of two or more different types of atoms and is called a gas. Some matter is made of two or more different types of atoms and is called a plasma.</p> </div> <p>Answer the questions that best complete the statement.</p> <p>1. What are the following elements?</p> <p>2. Helium</p> <p>3. Chlorine</p> <p>4. Oxygen</p> <p>5. Mercury</p> <p>6. Bromine</p> <p>7. Hydrogen</p> <p>8. What is the difference between a homogeneous and a heterogeneous mixture?</p> <p>9. They have uniform composition.</p> <p>10. They do not have uniform composition.</p> <p>11. They contain different substances.</p> <p>12. They have uniform composition.</p> <p>13. The number of particles in the large homogeneous mixture</p> <p>14. None</p> <p>15. Heterogeneous mixture</p> <p>16. Homogeneous mixture</p> <p>17. None</p> <p>18. 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