

Type of Chemical Reactions POGIL

Go Now: Use the chart below to answer the questions that follow.

1. Synthesis Reaction	2. Decomposition Reaction
 $2\text{Na} + \text{Cl}_2 \rightarrow 2\text{NaCl}$	 $\text{H}_2\text{CO}_3 \rightarrow \text{H}_2\text{O} + \text{CO}_2$
3. Single-Displacement Reaction	4. Double-Displacement Reaction
 $\text{Zn} + 2\text{HCl} \rightarrow \text{ZnCl}_2 + \text{H}_2$	 $\text{NaCl} + \text{AgF} \rightarrow \text{NaF} + \text{AgCl}$

- Which of the four reactions represented in the chart above shows two reactants exchanging partners to form two new compounds as products? **Fill in:** Recall that the arrow in a chemical equation ALWAYS points to the products of the reaction.
- Which of the four reactions represented in the chart above shows one element taking the place of another in a compound?

Flinn Scientific Pogil Types Of Chemical Reactions

PT Brinkman



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