



Michael Schwartz, PhD
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labflow



Figure 4 Boiling water bath setup

Observation Step

You will prepare a calibration curve for your thermometer using distilled water. Before preparation of the data, the value of the boiling point must be adjusted for atmospheric pressure. Although both melting points and boiling points change with varying pressure, the dependence of the melting point on pressure is very small and negligible. However, the rate of change of the boiling point with pressure is large. Atmospheric pressure at sea level is about 1013 mbar. The boiling point of water at 1013 mbar is equal to 100°C. However, the lab is near sea level but atmospheric pressure will be slightly lower than 1013 mbar. During the graph, you will fit the atmospheric pressure will be slightly lower than 1013 mbar. The atmospheric pressure on your report sheet during process. Please remember to record the pressure for the boiling point. The pressure for the boiling point can be estimated by assuming 1013 mbar. The result can use that atmospheric pressure to below 1013 mbar. The

$$P_{\text{atmosphere}} = (1013 \text{ mbar}) - (\text{atmospheric pressure}) = 0.833 \frac{\text{mbar}}{\text{cm}} \quad (1.0)$$

In additional from 1013 mbar to get the boiling point of water at the process.

$$P_{\text{atmosphere}} = (1013 \text{ mbar}) - (22 \text{ mbar}) = 991 \text{ mbar. The correction}$$

$$\text{The boiling point at 1013 mbar would be } (100.0^\circ\text{C}) - (0.18^\circ\text{C}) = 99.82^\circ\text{C.}$$

28. Change the axes titles in the spreadsheet to the appropriate units. The x-axis is temperature in °C and the y-axis is boiling point in °C. Copy the data into the spreadsheet and plot it on the manual axis.

Click on the side of the graph and change it to your name. Please will be the name of your graph at the side of your name.



29. Print out the graph and attach it to the papers that need to be handed over work.

Hunter Custom Lab Manual

KJ Lindholm-Leary



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