

General Chemistry Homework Answer Key

Textbook: OpenStax Chemistry: Atoms First 2e

Chapter 3: Electronic Structure and Periodic Properties of Elements

TA Reference Answers with Step-by-Step Explanations

3.1 Electromagnetic Energy

- 2 – Define frequency, wavelength, and amplitude.

Answer:

Frequency (ν) is the number of wave cycles that pass a point per second, measured in hertz (Hz).

Wavelength (λ) is the distance between successive crests of a wave, measured in meters (m).

Amplitude is the height of the wave from the center line to the crest or trough, indicating intensity.

Stepwise Procedure:

1. Define each term conceptually.
2. Include appropriate units.
3. Clarify differences among the properties.

- 4 – What is the relationship between frequency and energy?

Answer:

The energy (E) of electromagnetic radiation is directly proportional to its frequency (ν): $E = h\nu$, where h is Planck's constant (6.626×10^{-34} J·s).

Stepwise Procedure:

1. State the energy-frequency equation.
2. Define h and ν .
3. Explain that higher frequency means higher energy.

3.2 The Bohr Model

- 8 – What are the key features of the Bohr model of the hydrogen atom?

Answer:

1. Electrons orbit the nucleus in fixed paths (energy levels).
2. Only certain orbits are allowed with quantized energies.
3. Electrons absorb or emit energy when they move between levels.

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Yan Bai



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electrochemistry in clear straightforward language for undergraduates and mature scientists who want to understand solutions Like its predecessor the new text presents the electrochemistry of solutions at the molecular level The Second Edition takes full advantage of the advances in microscopy computing power and industrial applications in the quarter century since the publication of the First Edition Such new techniques include scanning tunneling microscopy which enables us to see atoms on electrodes and new computers capable of molecular dynamics calculations that are used in arriving at experimental values Chapter 10 starts with a detailed description of what happens when light strikes semiconductor electrodes and splits water thus providing in hydrogen a clean fuel There have of course been revolutionary advances here since the First Edition was written The book also discusses electrochemical methods that may provide the most economical path to many new syntheses for example the synthesis of the textile nylon The broad area of the breakdown of material in moist air and its electrochemistry is taken up in the substantial Chapter 12 Another exciting topic covered is the evolution of energy conversion and storage which lie at the cutting edge of clean automobile development Chapter 14 presents from a fresh perspective a discussion of electrochemical mechanisms in Biology and Chapter 15 shows how new electrochemical approaches may potentially alleviate many environmental problems

Elementary Modern Chemistry Wilhelm Ostwald, Harry Wheeler Morse, 1909 *Introduction to Modern Organic Chemistry* Carl H. Snyder, 1973 **Systematic Organic Chemistry** William Murdoch Cumming, Isaac Vance Hopper, 1923 **Modern Chemistry Pure and Applied** Arthur James Hale, 1921 **Introduction to Modern Chemistry: Student Manual** Eugene Meyer, 1979 **Journal of the Society of Chemical Industry** Society of Chemical Industry (Great Britain), 1886 English Mechanic and Mirror of Science and Art, 1879 **Volume 1: Modern Electrochemistry** John O'M. Bockris, Amulya K.N. Reddy, 1998-06-30

This book had its nucleus in some lectures given by one of us J O M B in a course on electrochemistry to students of energy conversion at the University of Pennsylvania It was there that he met a number of people trained in chemistry physics biology metallurgy and materials science all of whom wanted to know something about electrochemistry The concept of writing a book about electrochemistry which could be understood by people with very varied backgrounds was thereby engendered The lectures were recorded and written up by Dr Klaus Muller as a 293 page manuscript At a later stage A K N R joined the effort it was decided to make a fresh start and to write a much more comprehensive text Of methods for direct energy conversion the electrochemical one is the most advanced and seems the most likely to become of considerable practical importance Thus conversion to electrochemically powered transportation systems appears to be an important step by means of which the difficulties of air pollution and the effects of an increasing concentration in the atmosphere of carbon dioxide may be met Corrosion is recognized as having an electrochemical basis The synthesis of nylon now contains an important electrochemical stage Some central biological mechanisms have been shown to take place by means of electrochemical reactions A number of American organizations have recently recommended greatly increased activity in training and

research in electrochemistry at universities in the United States **Western Druggist** ,1887 *The Chemical News and
Journal of Industrial Science* ,1921 *An Introduction to Modern Experimental Organic Chemistry* Royston M. Roberts,1974
English Mechanic and Mirror of Science ,1879 **English Mechanic and World of Science** ,1879

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