

## Mass-Mole Conversion Worksheet

Name \_\_\_\_\_

Period \_\_\_\_\_

Calculate the molecular mass for each of the following molecules:

1. KOH

2.  $N_2O_2$

3.  $Cu_2SO_4$

4.  $Sr_3(PO_4)_2$

Using Factor-Labeling, convert each of the following from grams to moles:

5. 15.0 g  $C_2H_6$

6. 140.0 g NaOH

7. 27.2 g  $H_2O$

8. 45.7 g  $CaCO_3$

Using Factor-Labeling, convert moles to grams in each of the following:

9. 1.5 moles  $NH_3$

10. 0.65 moles  $H_2SO_4$

Using Factor-Labeling, convert the following to moles:

11.  $3.01 \times 10^{23}$  atoms Na

12.  $2.41 \times 10^{24}$  molecules  $CO_2$

Using Factor-Labeling, convert the following to atoms or molecules:

13. 2.56 moles Ca

14. 0.75 moles  $AlCl_3$

# Mole Lab Student Guide Answer Key

**Julia Schneider**



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### **Table of Contents Mole Lab Student Guide Answer Key**

1. Understanding the eBook Mole Lab Student Guide Answer Key
  - The Rise of Digital Reading Mole Lab Student Guide Answer Key
  - Advantages of eBooks Over Traditional Books
2. Identifying Mole Lab Student Guide Answer Key
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Mole Lab Student Guide Answer Key
  - User-Friendly Interface
4. Exploring eBook Recommendations from Mole Lab Student Guide Answer Key
  - Personalized Recommendations
  - Mole Lab Student Guide Answer Key User Reviews and Ratings
  - Mole Lab Student Guide Answer Key and Bestseller Lists
5. Accessing Mole Lab Student Guide Answer Key Free and Paid eBooks

- Mole Lab Student Guide Answer Key Public Domain eBooks
  - Mole Lab Student Guide Answer Key eBook Subscription Services
  - Mole Lab Student Guide Answer Key Budget-Friendly Options
6. Navigating Mole Lab Student Guide Answer Key eBook Formats
- ePub, PDF, MOBI, and More
  - Mole Lab Student Guide Answer Key Compatibility with Devices
  - Mole Lab Student Guide Answer Key Enhanced eBook Features
7. Enhancing Your Reading Experience
- Adjustable Fonts and Text Sizes of Mole Lab Student Guide Answer Key
  - Highlighting and Note-Taking Mole Lab Student Guide Answer Key
  - Interactive Elements Mole Lab Student Guide Answer Key
8. Staying Engaged with Mole Lab Student Guide Answer Key
- Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Mole Lab Student Guide Answer Key
9. Balancing eBooks and Physical Books Mole Lab Student Guide Answer Key
- Benefits of a Digital Library
  - Creating a Diverse Reading Collection Mole Lab Student Guide Answer Key
10. Overcoming Reading Challenges
- Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
11. Cultivating a Reading Routine Mole Lab Student Guide Answer Key
- Setting Reading Goals Mole Lab Student Guide Answer Key
  - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Mole Lab Student Guide Answer Key
- Fact-Checking eBook Content of Mole Lab Student Guide Answer Key
  - Distinguishing Credible Sources
13. Promoting Lifelong Learning
- Utilizing eBooks for Skill Development

- Exploring Educational eBooks
14. Embracing eBook Trends
- Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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