

# Chemistry Midterm Study Guide

**Periodic Table of the Elements**

**Legend:**

- Alkali Metals
- Alkaline Earth Metals
- Transition Metals
- Halogens
- Noble Gases
- Lanthanides
- Actinides

**Electronegativity:** Increases from left to right and from top to bottom.

**Atomic Radius:** Increases from right to left and from top to bottom.

- Period** - is horizontally row of elements in the periodic table  
ex - Chlorine has period 3rd and atomic number, while Fluorine has 2nd
- Group** - is vertically row of elements in the periodic table  
ex - Fluorine has period 2nd and atomic number, Chlorine has 3rd and atomic number
- Electronegativity** - atom's ability that how low electron will be attracted without a competing like ability of the nucleus  
ex - Fluorine has highest electronegativity, while Francium has lowest
- Atomic Radius** - The ability or tendency of an element to combine with other atoms like how far away the electrons are  
ex - Francium has highest atomic radius, Helium has lowest

# High School Chemistry Midterm Study Guide

**Padhraic Smyth**



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