

General Chemistry Semester 2 Study Guide

Mr. Pitzer - Fall 2018

1st Period: Monday, June 11th, 8:00am - 10:00am

2nd Period: Wednesday, June 13th, 11:00am - 1:00pm

3rd Period: Thursday, June 14th, 11:00am - 1:00pm

Use your resources wisely throughout the semester. You may not get to the end of the book because of a variety of circumstances that are beyond your control. Nevertheless, make every effort to complete the year of chemistry that you are enrolled in, please, and I would appreciate your feedback. Please contact your teacher, Mr. Pitzer, for additional insight.

Make yourself a learning schedule of the material covered throughout the semester. This is not just about what you suggest, but a more consistent & rigorous schedule of your study that will make your studying more efficient. Use the completed schedule to identify whether you have more work that requires additional attention.

Be flexible, creative, and diligent in your study approach, suggestions, suggestions, and observations.

Study Guide: Reviewable

Below are your final five following semester material available and that you have already been permitted to study.

1. Periodic Table
2. Chemical Properties: Acids
3. Ionic Compounds: Charges
4. Stoichiometry: Molar Mass
5. Naming: Ionic, Molecular, and Organic Compounds (Carbon, Alkanes, Alkenes, Alkynes, Alcohols, Aldehydes)
6. Unit Conversions: Mass, Molar Mass, Molar Mass
7. Chemical Equations: Balancing Equations

Final Exam: Knowledge Challenge

Below are your final five following semester and reviewable material. Students suggest, observations, and feedback are needed on appropriate questions.

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| <ol style="list-style-type: none">1. Molecular Structure: Empirical Formulas (Molar Mass)2. Molecular Structure: Lewis Structures3. Molecular Structure: Molecular Geometry4. Molecular Structure: Bond Lengths5. Molecular Structure: Bond Angles6. Molecular Structure: Bond Polarity7. Molecular Structure: Dipole Moments8. Molecular Structure: Hybridization9. Molecular Structure: Molecular Orbital Theory | <ol style="list-style-type: none">1. Molecular Structure: Lewis Structures2. Molecular Structure: Molecular Geometry3. Molecular Structure: Bond Lengths4. Molecular Structure: Bond Angles5. Molecular Structure: Bond Polarity6. Molecular Structure: Dipole Moments7. Molecular Structure: Hybridization8. Molecular Structure: Molecular Orbital Theory |
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General Chemistry Semester 2 Study Guide

AW Rasmussen



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